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COORDINATE DIAGRAM FOR SALINE HORIZONS

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DEFINITION

Saline Horizon is defined by two chemical criteria:

- **SALINE CONCENTRATION** : Electrical Conductivity in Saturation Extract (25°C), must be at some time of the year and at least in 15 cm thickness:

$$\begin{aligned} \text{EC} &> 15 \text{ dS.m}^{-1} \text{ if } 3.5 < \text{pH} < 8.5 \\ \text{EC} &> 8 \text{ dS.m}^{-1} \text{ if } 3.5 > \text{pH} > 8.5 \end{aligned}$$

Remark : High saline concentration in soil solution leads obligatory to adsorption of Na or Mg on the exchangeable complex, especially in non calcic environments, but the structure remains stable.

- **SOLUBILITY** : The possible salts must be more soluble than gypsum ($\log K 25^{\circ}\text{C} > -4.85$). They are distinguished *inside of the anionic triangular diagram* Fig. 1 : Neutral (1, 2, 3), Acid (4), or Alkaline salts (5, 6).

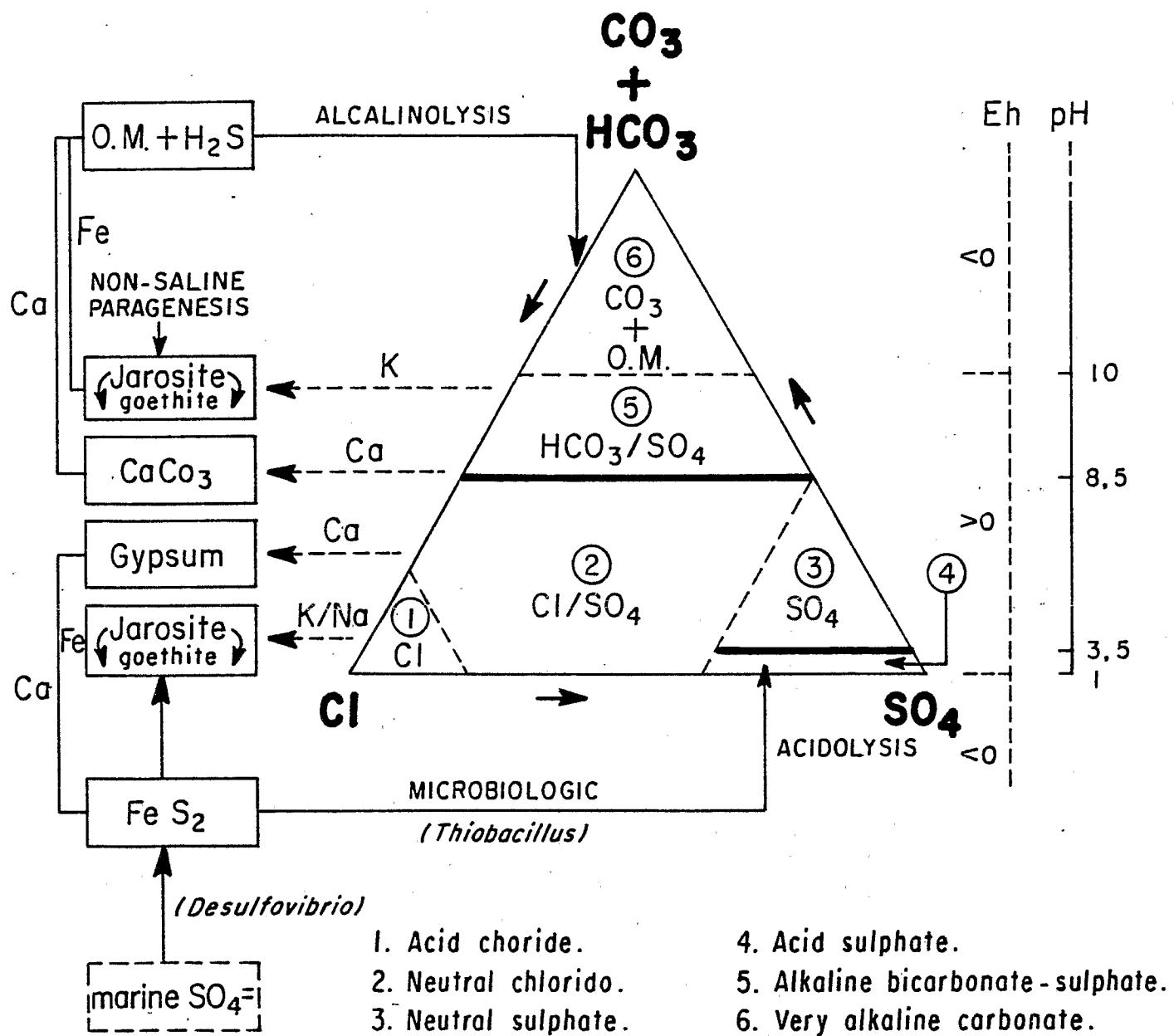
Within this EC range all the others present salts with a solubility product less than gypsum, are considered like non-saline mineral paragenesis (*outside of the triangle*).

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Cote B



Main salt-affected horizons.

EQUIVALENT TERMS

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Salic Horizon

- (1) Acid chloride
- (2) Neutral chlorido-sulfate
- (3) Neutral sulfate
- (4) Acid sulfate
- (5) Alkaline sulfate-bicarbonate
- (6) Very alkaline carbonate

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Saline Horizon

- Chloron
- Halon
- Sulphon
- Thion
- Alkalon
- Alkalon

FRENCH REFERENTIAL

Salique Horizon

- Neutral ($EC > 15 \text{ dS.m}^{-1}$
and $pH < 8.5$) :
Chlorido-sulfate / Sulfate.
- Alkaline ($EC > 8 \text{ dS.m}^{-1}$
and $pH > 8.5$) :
Bicarbonate / Carbonate.

U.S.D.A.

Salic Horizon

Thickness $> 15 \text{ cm}$ with
in weight $> 2\%$ salts more
soluble in cold water than
gypsum. Product of thickness
and salt % $>$ or = 60.
(without anionic divisions)

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