# Phenolic acids from Imperata cylindrica (L.) Raeusch. var. Major (Nees) c.e. Hubb.

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## INTRODUCTION

Alang-alang rhizomes, Imperata cylindrica (L.) Raeusch. var. major (Nees) C.E. Hubb., is widely used traditionally as a diuretic, antipyretic, anti-inflammatory, and haemostatic agent<sup>1, 2, 3, 4</sup>.

Alang-alang is know by various local names e.g. eurih (West Java), langalang, alang-alang, kambengan (Central and East Jabva), kebut, lalang (Madura)<sup>1, 2</sup>.

Imperata cylindrica var. major is an annual grass with long creeping, branched, scaly rhizomes; stem of vertical, simple (flowering) shoots not elongating and remaining underground until shortly before anthesis; basal leaves vigorous, with coriaceous sheaths, blades erect, linear-lanceolate, gradually attenuate downwards, upwards gradually tapering into acute, usually hard and sharp point, inflorescence a silky panicle, fruit caryopsis<sup>5, 6</sup>.

For man, the *Poaceal Gramineæ* is the one of the most important family in the plant kingdom because it is a source for food, building material etc., so that most of the chemical investigations are directed to compounds which are important economically *e.g.* carbohydrates, proteins, vitamins, volatile oils, while the secondary metabolites are still relatively unknown<sup>7</sup>.

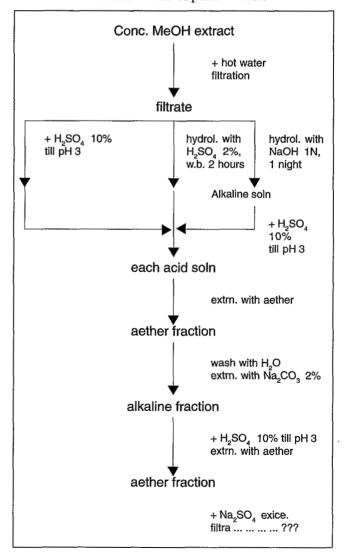
To support the governmental program in promoting the use of traditional drugs, an investigation on the chemical contents of *alang-alang* rhizomes, especially an analysis of the phenolic acids, was carried out.

# MATERIALS AND METHODS

The methanol extract of *alang-alang* rhizomes was obtained by soxhletation, evaporated under pressure at a temperature below 60 °C using a rotary evaporator.

Free phenolic acids, phenolic acids as glycosides and esters were investigated. The isolation schema of the phenolic acids can be seen in Fig. 1.

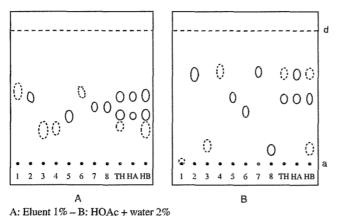
Fig. 1
Isolation scheme of phenolic acids



Identification of the phenolic acids is done by paper chromatography on Whatman 1 paper, with HOAc-water 2% and benzene-HOAc-water (60:22:1,2) as eluents, together with the reference standards.

For detection of the acids, the papers are sprayed with a diazoparanitroaniline solution, followed by a 15% sodium carbonate solution<sup>8</sup> (Fig. 2).

Fig. 2
Paper chromatography of phenolic acids on Whatman 1



Analysis of the phenolic acid components is done directly using the methanol extract and after acid and alkaline hydrolysis of the methanol extract, using two-dimensional paper chromatography,

with the same eluents mentioned before (Fig. 3, 4, 5).

Isolation of the various phenolic acids is done with preparative paper chromatography on Whatman 3 paper.

Identification of each phenolic acid is confirmed by ultraviolet spectral measurements and comparing with the ultraviolet spectra of the reference standards (Fig. 1.).

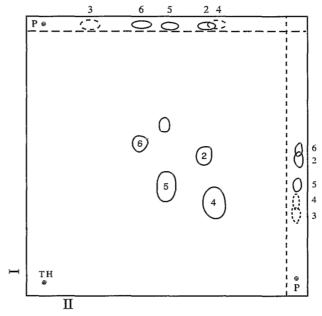
## RESULTS AND DISCUSSION

According to the results of the two-dimensional paper chromatography together with the reference standards, from the position, the fluorescence under UV-light and the colour of the spots after spraying, it can be concluded that:

- the methanol extract (before hydrolysis) contains vanillic-, ferulic-, p-coumaric-, and p-hydroxybenzoic acids.
- the methanol extract after acid hydrolysis contains vanillic,
   p-coumaric-, and p-hydroxybenzoic acids.
- the methanol extract after alkaline hydrolysis contains vanillic-, p-coumaric-, p-hydroxybenzoic and caffeic acids.

The ultraviolet spectra of vanillic-, ferulic-, p-coumaric, and caffeic acids show differences with the ultraviolet spectra of the reference standards, because the isolated phenolic acids,

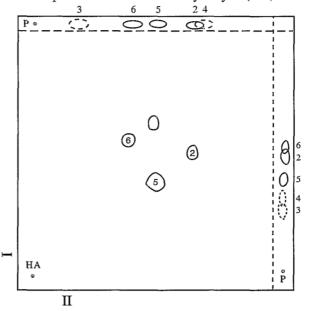
Fig. 3
Two-dimensional paper chromatography
of free phenolic acids (TH)



I.: Eluent HOAc-water 2% – II.: Eluent benzene-HOAc-water (60:22:1,2) Relative position and colour of spot 2 (violet), 4 (blue-brown, blue fluor.), 5 (blue-brown), and 6 (rose) corresponds with vanillic-, ferulic-, p-coumaric, and p-hydroxybenzoic acid.

Fig. 4

Two-dimensional paper chromatography of phenolic acids after acid hydrolysis (HA)

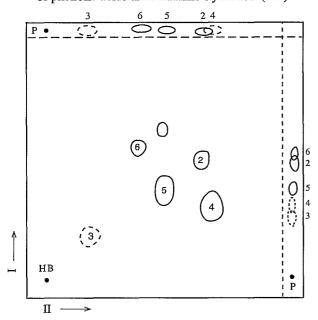


I.: Eluent HOAc-water 2% – II.: Eluent benzene-HOAc-water (60:22:1,2) Relative position and colour of spot 2 (violet), 5 (blue-brown), and 6 (rose) corresponds with vanillic-, p-coumaric, and p-hydroxybenzoic acid.

especially ferulic-, p-coumaric, and caffeic acids, are always obtained as mixtures of *cis*- and *trans*-isomers, while the isolated vanillic acid is probably still mixed with the *cis*-isomer of ferulic acid<sup>9</sup>, p-Hydroxybenzoic acid cannot be obtained in the pure form because it is always mixed with the *cis*-isomer of p-coumaric acid, and due to its very small amount no ultraviolet spectrum could be made.

Fig. 5

Two-dimensional paper chromatography of phenolic acids after alkaline-hydrolisis (HB)



I.: Eluent HOAc-water 2% – II.: Eluent benzene-HOAc-water (60:22:1,2) Relative position and colour of spot 2 (violet), " (brown, blue fluor.), 4 (blue-brown, blue fluor.), 5 (blue-brown), and 6 (rose) corresponds with vanillic-, caffeic-, ferulic-, p-coumaric, and p-hydroxybenzoic acid.

## CONCLUSION

From the results it can be concluded that *alang-alang* rhizomes contain vanillic-, ferulic-, p-coumaric-, and p-hydroxybenzoic acids in the free form, and caffeic acid in the ester form.

## REFERENCES

- 1. HEYNE K., 1953, *De nuttige planten van Indonesië*, Deel I, 3de Ed., Bandung, W. Van Hoeve, 153-158.
- 2. DITJEN P.O.M., DEPKES R.I., 1979, Materia Medika Indonesia, Vol. III, Jakarta, Depkes R.I., 83-87.
- 3. PERRY L.M., 1978, Medicinal plants of East and Southeast Asia, London, The Mit Press, 166-167.
- 4. PONGLUX D. et al., 1987, Medicinal Plants, Bangkok, Victory Power Point Corp., 149.
- 5. BACKER C.A., BAKHUIZEN VAN DEN BRINK R.C., 1968, Flora of Java, Vol. III, Groningen, Wolters-Noordhoff N.V., 501-583.
- 6. VAN STEENIS C.G.G.J., 1975, Flora untuk sekolah di Indonesia, Jakarta, P.T. Pradnya Paramita, 107-108.
- 7. HEGNAUER R., 1963, *Chemotaxonomie der Pflanzen*, Band 2, Basel, Birkhäuser Verlag, 156-227.
- 8. HARBORNE J.B., 1973, *Phytochemical methods*, London, Chapman And Hall, 34-50.
- 9. FRY S.C., 1988, The growing plant cell wall: chemical and metabolic analysis, Essex, Longman Scientific and Technical, 62, 160.