

Ore Texture, Mineralogy and Whole Rock Geochemistry of the Iron Mineralization from Edea North Area, Nyong Complex, Southern Cameroon: Implication for Origin and Enrichment Process

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Abstract

This study presents the ore texture, mineralogy and whole rock geochemistry of iron occurrence from the Edea North area located at the upper limit of the Nyong Unit, part of the Congo craton. This iron mineralization is of magnetite quartzite type enclosed by metamorphic rocks of the granulite facies. Two main facies have been identified over the study area including the banded and the massive facies. The mineralogical set of these facies is in majority represented by magnetite, quartz and pyroxene martite in addition to minor biotite, apatite, and amphibole. Magnetite presents as irregular and elongated minerals which can contain quartz inclusions of various shapes and sizes, as blasts clustering, around pyroxenes and isolated xenomorphic magnetite minerals in the silicate phases. Quartz varies from fine to coarse-grained and ribbon quartz. This indicates re-crystallization and deformation during metamorphism. Electron microprobe analysis on magnetite and martite show Fe₂O₃ contents that vary between 96.11 and 99.76. Whole rock chemical data showed that Fe-contents are as high as 62.9 wt%. The SiO₂ content varies between 33.8 wt% and 51.2 wt%. Iron oxides and SiO₂ are negatively correlated. Moreover, the low positive correlations between Al₂O₃ and HFSE, and LILE suggest a contribution of clastic materials in the protolith of studied materials. The samples show low contents of V, Cr, Ba, Zr with respect to igneous rocks. This may infer a sedimentary origin for the studied rocks; furthermore, these materials may have undergone hydrothermal alteration. The REE patterns re-

veal enrichment in LREE compared to HREE. Some samples show positive Eu (1.82) and other negative anomalies Eu (0.54 to 0.97). The lack of Ce anomalies suggests that the depositional environment of the Edea North iron occurrence was anoxic or suboxic. Overall, the Edea North iron occurrence may have undergone significant input from hydrothermal sources with imprints of the clastic during its deposition.

Keywords

Magnetite, Texture, Edea North, Nyong Complex, Southern Cameroon

1. Introduction

Low iron commodity prices since January 2014 acted for low expenditure on iron exploration and dearth of discoveries of new iron deposits around the world. However, iron, being a raw material of steel, is a key issue for the growing global steel industry. Thus, despite the current low price of iron, the long-term trend for iron is positive. Classically, there are two types of Precambrian iron formations discriminated based on their depositional setting: 1) Algoma-type iron formations are in close proximity to ancient volcanic centers suggesting a sub-aqueous hydrothermal origin similar to modern day sea-floor spreading centers [1]; 2) the Lake Superior-type BIFs are developed in passive-margin sedimentary rock successions and generally lack direct relationships with extrusive volcanic materials and are therefore interpreted as chemical precipitates of iron-rich waters in a shallow sea [2]. Recently, [3] have proposed the composition-based definition of Fe-formations as “siliceous and Fe-rich sedimentary chemical precipitates with low levels of detrital siliciclastic or volcanoclastic material (<1% Al_2O_3) and greater than 10% total Fe, regardless of whether Fe is associated with a carbonate or oxide phase”. This definition of Fe formation encompasses both granular and banded Fe formations as well as ferruginous cherts (e.g. distal hydrothermal jaspers). Moreover, the quality of iron ore is a key issue in the steel industry. The increasing global demand for iron, however, has made BIF-hosted low grade iron ore (Fe < 40 wt.%) important targets for exploration. In addition to the increasing scarcity of economic hematite deposits worldwide lead magnetite to emerge as an important source of the world's iron.

Although Cameroun is not yet an iron producer, past works and particularly within the ongoing decade have led to the discovery in Precambrian greenstones terrain of south Cameroon, of several BIF-hosted iron deposits including Mbalam [4] [5] [6] [7], Nkout [8] [9], Bikoula (www.aluvance.com), Elom [10], and Zambi deposits [11] and Kouambo deposit [12]. Most of these studies have used BIF to infer the origin and depositional environment of these iron formations. Although the BIFs are less widely distributed than other lithologies (e.g., Archean TTG), they are an integral and unique part of the Ntem Complex which corresponds to the northern edge of the Congo craton [12]. However, BIF are

not the only iron formation in this complex. Recent works by the “Compagnie Minière du Cameroun” (CMC) (see www.africaminerals.com) have evidenced magnetite gneisses with estimated reserves of 10 millions of tons in the Edéa North iron, within the Nyong unit which is part of the Archaean Ntem complex. However, no scientific work on these formations is available in the literature.

The present work was conducted at the time of renewed interest in the iron ore potential of the Edéa North iron ore project. This article presents ore texture, mineralogy and whole rock geochemistry of the Edéa North iron occurrence with the aim at characterizing these iron-formations as well as to infer their enrichment processes.

2. Geological Setting

The Archaean Ntem Complex greenstones belt is located within the northern margin of the Congo craton in Cameroon [13] [14] [15]. This complex is made up of Archaean and Palaeoproterozoic rocks and constitutes a relative stable block limited in the north by the Pan-African orogenic belt (e.g. Yaounde Group) [16] [17]. The Ntem Complex greenstones belt trends E-W and extends over 500 km from Mbalam in the South East to Kribi in Atlantic coast in the West. This belt has been subdivided into three main units including the Ayna Unit, the Ntem Unit and the Nyong Unit respectively from the East to the West. The Ntem complex was affected by two major periods of deformation. The first involves successive diapiric emplacements of the Mesoarchean charnockites (~2.900 Ma) and TTGs (~2830 Ma). It is marked by vertical foliation and lineation, stretching and isoclinal folds [18]. This episode of deformation was synchronous with a regional granulite-facies metamorphism [19]. This Archaean event was followed, in the Paleoproterozoic, by a trans-current deformation phase marked by the development of N-S to NE-SW trending sinistral shear zones and partial melting of the TTG suite and the greenstone belt country rocks, with generation of a variety of granites [20]. Late syenitic plutons (~2.3 Ga) intruded the complex during this second tectonic episode [21]. The Eburnean metamorphism affected all the Archaean and Paleoproterozoic formations and was dated at ~2.05 Ga [22]. The Edéa North area is located at the upper limit of the Nyong Unit (**Figure 1(a)**). This Unit is dominated by biotite-hornblende gneisses, which locally appear as grey gneisses of TTG composition, orthopyroxene-garnet gneisses (charnockites), garnet amphibole-pyroxenites, and banded iron-formation [11] [13] [23] [24]. This Unit is of Mesoarchaean to Palaeoproterozoic [23], and belongs to the West Central African Belt (WCAB) [25] [26]. The study site is composed of the charnockitic suit rocks, orthogneisses, mylonites and magnetite gneisses (**Figure 1(b)**).

3. Sampling and Analytical Methods

A total of 8 iron mineralization sample were selected on the base of their massive appearance and magnetic properties and prepared for microscopic petrography and geochemical analyses. Polished thin sections were prepared at the “Laboratoire

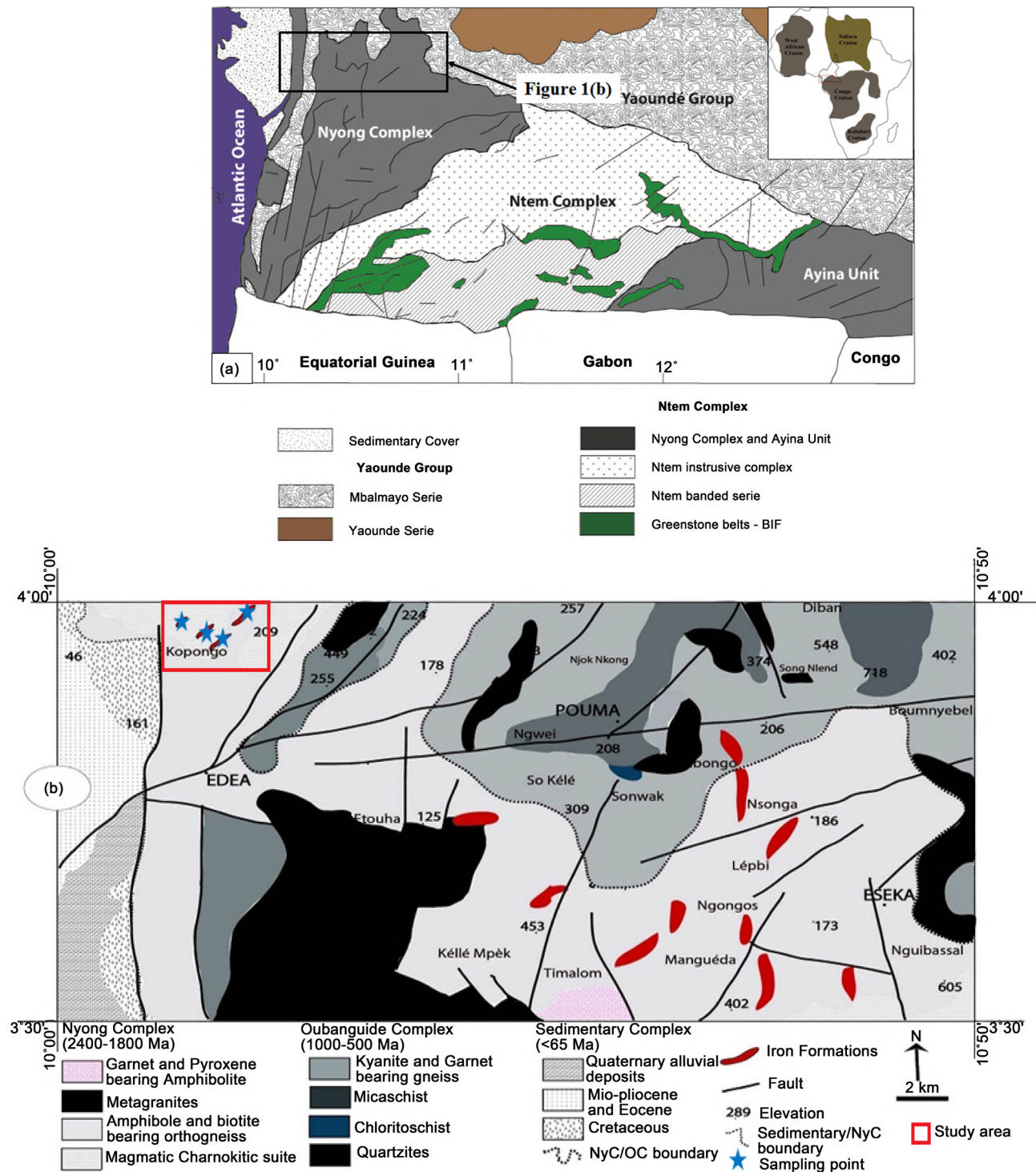


Figure 1. Location of study area. (a) Simplified geological map of South-West Cameroun [13]; (b) Geological map of Edea-Eseka modified after [27].

de propriétés des amorphes, liquides et minéraux de l'institut de physique des matériaux et de cosmochimie de l'Université de Pierre et Marie Curie (France)" for petrography and EMPA. EMPA on iron minerals was achieved in the up-mentioned laboratory. The samples were analyzed for Al_2O_3 , Fe_2O_3 , SiO_2 , TiO_2 , Cr_2O_3 and MnO using an acceleration voltage of 15 KeV and a beam cur-

rent of 150 nA. Natural and artificial silicates and oxides were used as standards. The detection limit was set at 0.01 wt% for the entire major elements and relative measurement errors not exceed 1%. Petrographic observations of the polished sections was done using both transmitted and reflected light at the Economic Geology Unit laboratory, University of Buea, Cameroon. Whole rock geochemical analyses were performed at ALS Minerals Global Group in South Africa and Canada using Inductively Coupled Plasma-Atomic Emission Spectrometry (ICP-AES) for major elements and Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) for trace and Rare Earth Elements. During analysis, 0.2 g of the rock powder was fused with lithium or lithium metaborate/tetraborate (LiBO_2) and dissolve in nitric acid (HNO_3). Loss on ignition (LOI) was determined by firing the powder at 1000°C and calculating the weight difference before and after the firing. In order to maintain the data quality, various standards and unknowns were run between the analyses.

4. Results

4.1. Textures and Ore Mineralogy

Hand specimens of iron mineralization samples from the Edea North were divided into two main facies based on their physical appearance: the banded and the massive ore (**Figure 2**). Bands in the banded facies are irregular and discontinuous there are made up of thin bands of ferromagnesian minerals which alternate with silica-rich microbands and the contact banding is not sharp (**Figure 2(a)**). Some of the studied samples displayed a trellis-like structure (**Figure 3**). Microscopic examinations of these samples have confirmed the irregularity of the bands. The mineralogy of the banded samples consists of iron minerals (75%), pyroxenes (25%) in the ferromagnesian bands and quartz (80%), and dominating silica-rich bands. The mineralogical composition of the massive ore includes iron minerals (45%), quartz (30%) and pyroxene (16%) with minor, ilmenite (1%), biotite (0.5%), apatite (0.5%) and amphiboles (1%). The reflected light microscopic observations revealed that the iron minerals phase is dominated by magnetite in both facies and to a lesser amount martite. All the pyro-

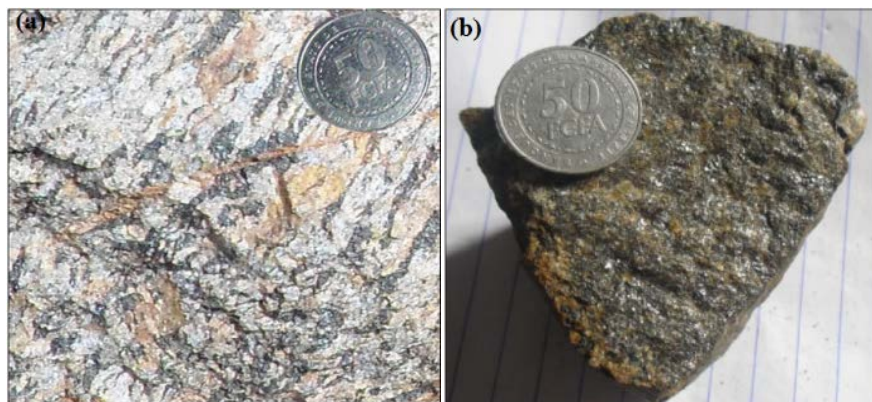


Figure 2. Representative iron ores from Edea North iron mineralization. (a) Banded facies. Note the irregular band (b) Massive ore facies.

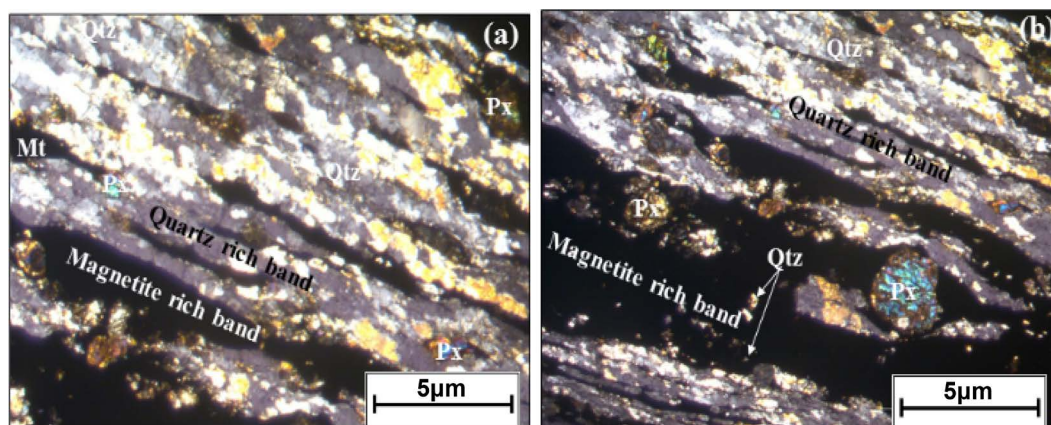


Figure 3. Representative photomicrographs of iron ores from the Edea north. A weak banding with alternating magnetite and quartz bands can be identified. (a) Discontinuous magnetite bands with quartz inclusions. (b) The embedded pyroxene as inclusion in magnetite ribbon. Mt = magnetite, Qtz = quartz, Px = pyroxene.

xenes minerals present marks of alteration and some of them are deformed within the bands (**Figure 3(b)**).

Magnetite in the both facies occurs as in various shapes: as irregular and elongated minerals can contain quartz inclusions of various shapes and sizes (**Figure 3(a)**). Magnetite also appears as cluster around pyroxenes and embedded at their limits, thus forming haloes and rolls (**Figure 4(c)** and **Figure 4(d)**). Finally magnetite occurs as disseminations in quartz matrix (**Figure 4(c)** and **Figure 4(d)**). Martite results from the alteration of magnetite is altered to martite (**Figure 5**) and rare inclusions of ilmenite occur within. Martite is euhedral to subhedral and it is commonly observed at the crystal interface of magnetite and silicate strip.

Quartz is the main gangue mineral identified. It displays three textural varieties including fine-grained, coarse-grained and ribbon quartz (**Figure 3** and **Figure 4**). It occurs in band or as inclusions in magnetite or pyroxene. In addition, quartz shows a characteristic undulating texture (**Figure 4(b)**). The observed crystals are fractured, xenomorphic and subhedral to anhedral phases.

The pyroxenes occur as inclusions in both silica and magnetite. They are subhedral, yellow-orange, and riding in thin strips of magnetite forming rolls (**Figure 3(b)**) and always close or in the iron mineral phase.

Amphiboles are sometimes green xenomorphic crystals with quartz inclusions.

4.2. Mineral and Whole Rock Geochemistry

Electron microprobe analysis revealed that $\text{Fe}_2\text{O}_{3(\text{T})}$ contents range from 97.12% to 99.76% and between 96.11% and 97% respectively on magnetite and martite (**Table 1**). All other major oxides occur in low quantities except Al_2O_3 with contents as high as 2.65%.

The chemical composition of iron ore from the Edea North area is summarized in **Table 2**. The samples show SiO_2 contents that range from 33.8 to 51.2

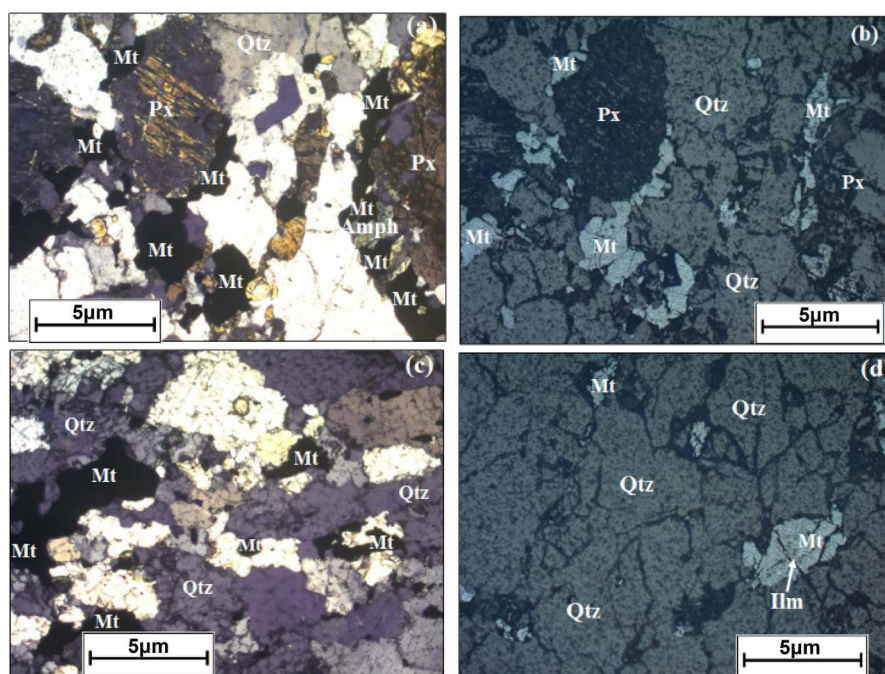


Figure 4. Representative polarized and reflected ((c), (d)) light photomicrographs and showing the various magnetite grained texture identified in the Edea north iron mineralization. (a) and (b) Magnetite cluster texture around pyroxene, (b) and (c) Magnetite isolated crystals textures. Note the undulating and coarse-grained textures of quartz. Mt: magnetite; Mrt: martite; Amph: Amphibole; Px: pyroxene; Qtz: quartz; Ilm: Ilmenite.

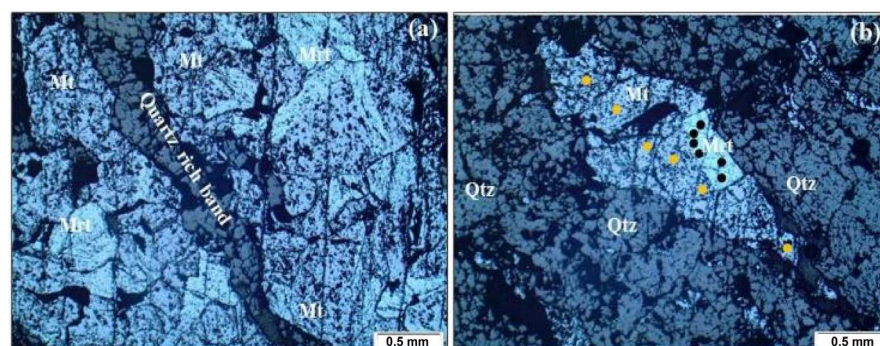


Figure 5. Representative reflected light photomicrographs showing the martite minerals in the iron mineralization in the Edea north iron occurrence. The yellow and black circle represents the EPMA analysis point on Magnetite and Martite respectively.

wt% and Fe_2O_3 contents that vary between 40.4 and 62.9 wt% respectively. There is a very strong negative correlation ($R^2 = -0.92$) between these two oxides (Table 3). Concentrations of Al_2O_3 vary between 0.46% and 4.19% (with an average of 1.40%) and P_2O_5 concentrations are generally low (0.01% - 0.15%). The other major oxides have relatively low contents: CaO (0.01% - 0.47%), K_2O (0.01% - 0.02%), Na_2O (0.01% - 0.08%) and MnO (0.01% - 0.09%). there is a positive correlation ($R^2 = 0.89$) between Al_2O_3 and TiO_2 while Al_2O_3 and Fe_2O_3 are negatively correlated ($R^2 = -0.60$) (Table 3).

Trace elements are weakly concentrated in the studied samples. The High Field Strength Elements (HFSE), Hf, Zr, Th and U vary from 0.2 - 0.4 ppm, 7 -

37 ppm, 0.19 - 15.5 ppm and 0.09 - 0.72 ppm respectively (**Table 2**). The contents of Large Ion Lithophile Elements (LILE), Sr, Rb, and Ba vary from 1 to 27.3 ppm, 0.2 to 1.7 ppm and 1.6 to 131 ppm respectively. The studied iron ore are depleted in HFSE and LILE when compare to the upper continental crust after [28].

Rare earth elements concentration in the iron ore from the Edea North area vary with Σ REE oscillating between 5.31 and 397.74 ppm (average of 80.41) (**Table 2**). The samples are enriched in Light Rare Earth Elements (LREE) compared to Heavy Rare Earth Elements (HREE) with LREE/HREE ratios between 10.35 and 55.61 (average of 24.37) (**Figure 6**). The samples show a negative Eu anomaly (Eu/Eu^* between 0.54 and 0.97) except EDN19 and EDN17 that reveal a slightly positive Eu anomaly (Eu/Eu^* values range from 1 to 1.83). Ce displays negative anomalies between 0.39 and 0.87. The majority of the samples showed a negative anomaly in La and none in Ce and one sample displayed a positive anomaly in Ce (**Figure 7**).

5. Discussion

5.1. Mineralogical Association and Nature of the Edea North Iron Occurrence

Petrographic examination of the studied samples shows the following mineral assemblage: Magnetite + Quartz + Pyroxene \pm Biotite \pm Martite \pm Amphibole. This mineralogy is characteristic of iron formations that have undergone metamorphism in the granulite facies [29]. Quartz and magnetite-martite identified in the samples is consistent with the geochemistry that shows Fe_2O_3 contents from 40.4 wt% to 62.9 wt% and SiO_2 contents between 33.8 wt% and 51.2 wt%.

Table 1. Electron microprobe analyses on magnetite and martite from the Edéa North iron mineralization.

| Minerals | Data set/point | Al_2O_3 | SiO_2 | TiO_2 | Cr_2O_3 | MnO | Fe_2O_3 | Total |
|------------------|----------------|-------------------------|----------------|----------------|-------------------------|------|-------------------------|--------|
| <i>Magnetite</i> | 1 | 0.25 | 0.02 | 0.05 | bdl | 0.08 | 99.30 | 99.70 |
| | 2 | 0.17 | bdl | 0.03 | bdl | 0.05 | 99.76 | 100.00 |
| | 3 | 0.20 | 0.03 | 0.06 | bdl | 0.05 | 99.67 | 100.00 |
| | 4 | 2.65 | 0.02 | 0.20 | 0.001 | 0.02 | 97.12 | 100.00 |
| | 5 | 0.17 | 0.02 | 0.03 | bdl | 0.05 | 99.72 | 100.00 |
| | 6 | 0.94 | 0.02 | 0.15 | 0.001 | 0.06 | 98.83 | 100.00 |
| <i>Martite</i> | 1 | 0.17 | 0.03 | 0.03 | bdl | 0.04 | 96.11 | 96.38 |
| | 2 | 0.15 | 0.02 | 0.02 | bdl | 0.06 | 96.38 | 96.61 |
| | 3 | 0.16 | 0.02 | 0.03 | 0.00 | 0.05 | 96.94 | 97.20 |
| | 4 | 0.20 | 0.05 | 0.03 | 0.00 | 0.08 | 96.75 | 97.10 |
| | 5 | 0.17 | 0.02 | 0.02 | 0.00 | 0.05 | 97.00 | 97.26 |
| | 6 | 0.17 | 0.01 | 0.02 | bdl | 0.05 | 96.24 | 96.48 |

bdl = below detection limit

Table 2. Whole rock major, trace and rare earth element composition of samples of the iron mineralization from the Edea north, southern Cameroon.

| Sample ID | Banded iron ore facies | | | | Massive iron ore facies | | | | Avg. |
|--------------------------------|------------------------|--------|-------|--------|-------------------------|--------|-------|--------|--------|
| | EDN04 | EDN 12 | EDN18 | EDN19 | EDN09 | EDN14 | EDN15 | EDN17 | |
| SiO ₂ (wt%) | 44.1 | 42.8 | 35.4 | 48.1 | 45.9 | 51.2 | 35.1 | 33.8 | 42.05 |
| TiO ₂ | 0.1 | 0.16 | 0.11 | 0.06 | 0.05 | 0.36 | 0.06 | 0.35 | 0.16 |
| Al ₂ O ₃ | 0.47 | 1.38 | 1.29 | 0.64 | 0.63 | 4.19 | 0.46 | 2.14 | 1.4 |
| Fe ₂ O ₃ | 55 | 54.7 | 59.8 | 53.3 | 52 | 40.4 | 62.6 | 62.9 | 55.09 |
| MnO | 0.03 | 0.02 | 0.01 | 0.04 | 0.08 | 0.09 | 0.02 | 0.03 | 0.04 |
| MgO | 0.14 | 0.02 | 0.03 | 0.02 | 1.3 | 1.58 | 0.04 | 0.46 | 0.45 |
| CaO | 0.03 | 0.01 | 0.01 | 0.01 | 0.2 | 0.47 | 0.01 | 0.2 | 0.12 |
| Na ₂ O | 0.01 | 0.01 | 0.01 | 0.01 | 0.02 | 0.08 | 0.01 | 0.01 | 0.02 |
| K ₂ O | 0.01 | 0.01 | 0.01 | 0.01 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 |
| P ₂ O ₅ | 0.05 | 0.06 | 0.06 | 0.01 | 0.11 | 0.11 | 0.06 | 0.15 | 0.08 |
| LOI | 0.59 | 0.77 | 1.68 | -0.65 | -1.24 | 2.15 | 1.5 | 0.1 | 0.61 |
| Total | 100.53 | 99.94 | 98.41 | 101.55 | 99.07 | 100.64 | 99.87 | 100.15 | 100.02 |
| V (ppm) | 7 | 13 | 22 | 9 | 5 | 69 | 8 | 31 | 24.00 |
| Cr | 20 | 30 | 20 | 10 | 30 | 80 | 40 | 10 | 31.67 |
| Cs | 0.02 | 0.01 | 0.1 | 0.11 | 0.04 | 0.16 | <0.01 | 0.26 | 0.13 |
| Rb | 0.5 | 0.5 | <0.2 | <0.2 | 0.6 | 0.7 | <0.2 | 1.7 | 1.00 |
| Sr | 2.1 | 12.6 | 3.9 | 2.2 | 5 | 27.3 | 1 | 11.6 | 8.50 |
| Y | 1.9 | 5 | 2.8 | <0.5 | 3.8 | 109.5 | 1.2 | 8.2 | 25.10 |
| Zr | 10 | 13 | 11 | 14 | 8 | 37 | 11 | 7 | 14.67 |
| Ba | 6.1 | 74.9 | 39 | 2.2 | 12.7 | 131 | 1.6 | 120.5 | 51.17 |
| Th | 0.31 | 1.79 | 0.39 | 0.19 | 1.27 | 0.24 | 1.57 | 15.5 | 3.19 |
| U | 0.11 | 0.72 | 0.09 | 0.15 | 0.64 | 0.18 | 0.25 | 0.53 | 0.31 |
| Ta | 0.1 | 0.3 | 0.5 | 0.6 | 0.1 | 0.6 | 0.2 | 0.5 | 0.42 |
| Nb | 1.3 | 4.1 | 1.8 | 6.3 | 1.7 | 4.2 | 1.8 | 6.1 | 3.65 |
| Ga | 6.5 | 12.4 | 6.4 | 8.5 | 4.9 | 11.7 | 3.1 | 25.2 | 9.97 |
| Hf | 0.3 | 0.4 | 0.3 | 0.3 | 0.2 | 1.1 | 0.3 | 0.2 | 0.40 |
| Sn | 2 | 8 | 1 | 5 | 4 | 7 | 2 | 6 | 4.17 |
| W | <1 | 1 | 1 | <1 | 1 | <1 | <1 | 1 | 1.00 |
| Th/U | 2.82 | 2.49 | 4.33 | 1.27 | 1.98 | 1.33 | 6.28 | 29.25 | 7.41 |
| La | 2.9 | 5.2 | 2.9 | 0.7 | 4.9 | 46.9 | 0.9 | 53.1 | 18.23 |
| Ce | 5.4 | 11.7 | 5.7 | 1.3 | 10.8 | 160 | 2 | 39.6 | 36.57 |
| Pr | 0.73 | 1.45 | 0.76 | 0.11 | 1.2 | 17.7 | 0.24 | 10.4 | 5.07 |
| Nd | 2.5 | 6.3 | 3 | 0.6 | 4.8 | 77.8 | 1 | 33.7 | 20.15 |
| Sm | 0.48 | 1.44 | 0.45 | 0.09 | 0.87 | 20.3 | 0.23 | 6.23 | 4.70 |
| Eu | 0.08 | 0.25 | 0.17 | 0.04 | 0.22 | 4.21 | 0.05 | 1.57 | 1.04 |
| Gd | 0.42 | 1.3 | 0.73 | 0.05 | 0.89 | 19.55 | 0.23 | 3.72 | 4.20 |
| Tb | 0.05 | 0.17 | 0.06 | 0.01 | 0.1 | 3.17 | 0.03 | 0.48 | 0.64 |
| Dy | 0.39 | 0.9 | 0.47 | 0.05 | 0.61 | 19.5 | 0.23 | 2.62 | 3.91 |

Continued

| | | | | | | | | | |
|------------------------|-------|-------|-------|-------|-------|--------|-------|--------|--------|
| Ho | 0.08 | 0.17 | 0.08 | 0.01 | 0.12 | 3.94 | 0.05 | 0.44 | 0.77 |
| Er | 0.2 | 0.38 | 0.39 | 0.05 | 0.33 | 11.45 | 0.15 | 0.99 | 2.23 |
| Tm | 0.02 | 0.04 | 0.06 | 0.01 | 0.06 | 1.55 | 0.01 | 0.2 | 0.32 |
| Yb | 0.18 | 0.19 | 0.23 | 0.03 | 0.3 | 10.2 | 0.17 | 0.71 | 1.94 |
| Lu | 0.03 | 0.03 | 0.05 | 0.01 | 0.04 | 1.47 | 0.02 | 0.12 | 0.29 |
| ΣREE | 13.46 | 29.52 | 15.05 | 3.06 | 25.24 | 397.74 | 5.31 | 153.88 | 100.05 |
| LREE | 11.53 | 24.65 | 12.36 | 2.71 | 21.7 | 302.4 | 4.14 | 136.8 | 80.02 |
| HREE | 0.51 | 0.81 | 0.81 | 0.11 | 0.85 | 28.61 | 0.4 | 2.46 | 5.54 |
| LREE/HREE | 22.61 | 30.43 | 15.26 | 24.64 | 25.53 | 10.57 | 10.35 | 55.61 | 23.66 |
| (La/Yb) _{CN} | 10.04 | 17.99 | 6.02 | 7.27 | 12.72 | 3.31 | 4.67 | 45.94 | 13.32 |
| (Y/Y*) _{CN} | 1.04 | 1.05 | 0.68 | 0.72 | 0.97 | 0.95 | 0.97 | 0.99 | 0.88 |
| (Eu/Eu*) _{CN} | 0.54 | 0.56 | 0.91 | 1.82 | 0.76 | 0.65 | 0.66 | 1.00 | 0.97 |
| (Ce/Ce*) _{CN} | 0.87 | 1.00 | 0.90 | 1.10 | 1.04 | 1.30 | 1.01 | 0.39 | 0.96 |

$$(Eu/Eu^*)_{CN} = \left(\frac{Eu}{\sqrt{Sm*Gd}} \right)_{CN} ; (Ce/Ce^*)_{CN} = \left(\frac{Ce}{\sqrt{La*Pr}} \right)_{CN} ; (Y/Y^*) = \left(\frac{Y}{\sqrt{Dy*Ho}} \right)_{CN}$$

Table 3. Linear inter-elements correlations (R²) for all iron mineralization samples (n = 8).

| | SiO ₂ | Al ₂ O ₃ | Fe ₂ O ₃ | CaO | MgO | Na ₂ O | K ₂ O | TiO ₂ | MnO | P ₂ O ₅ | SrO | BaO | LOI | V | Cr | Rb | Sr | Y | Zr | Ba | Hf | Th | U | Th/U |
|--------------------------------|------------------|--------------------------------|--------------------------------|------|-------|-------------------|------------------|------------------|-------|-------------------------------|-------|-------|-------|-------|-------|------|-------|-------|------|-------|------|-------|------|------|
| SiO ₂ | 1 | | | | | | | | | | | | | | | | | | | | | | | |
| Al ₂ O ₃ | 0.30 | 1 | | | | | | | | | | | | | | | | | | | | | | |
| Fe ₂ O ₃ | -0.92 | -0.60 | 1 | | | | | | | | | | | | | | | | | | | | | |
| CaO | 0.45 | 0.86 | -0.70 | 1 | | | | | | | | | | | | | | | | | | | | |
| MgO | 0.54 | 0.63 | -0.72 | 0.92 | 1 | | | | | | | | | | | | | | | | | | | |
| Na ₂ O | 0.60 | 0.86 | -0.85 | 0.90 | 0.81 | 1 | | | | | | | | | | | | | | | | | | |
| K ₂ O | 0.24 | -0.25 | -0.17 | 0.20 | 0.54 | 0.00 | 1 | | | | | | | | | | | | | | | | | |
| TiO ₂ | 0.02 | 0.89 | -0.30 | 0.74 | 0.46 | 0.60 | -0.34 | 1 | | | | | | | | | | | | | | | | |
| MnO | 0.74 | 0.51 | -0.82 | 0.84 | 0.95 | 0.78 | 0.55 | 0.30 | 1 | | | | | | | | | | | | | | | |
| P ₂ O ₅ | -0.21 | 0.55 | -0.04 | 0.69 | 0.64 | 0.36 | 0.31 | 0.70 | 0.41 | 1 | | | | | | | | | | | | | | |
| SrO | -0.51 | 0.24 | 0.44 | 0.20 | 0.01 | -0.16 | -0.14 | 0.61 | -0.14 | 0.68 | 1 | | | | | | | | | | | | | |
| BaO | 0.56 | 0.89 | -0.82 | 0.86 | 0.72 | 0.99 | -0.14 | 0.64 | 0.69 | 0.31 | -0.14 | 1 | | | | | | | | | | | | |
| LOI | -0.18 | 0.52 | -0.12 | 0.19 | -0.03 | 0.44 | -0.64 | 0.38 | -0.19 | 0.00 | -0.18 | 0.53 | 1 | | | | | | | | | | | |
| V | 0.28 | 0.99 | -0.59 | 0.85 | 0.61 | 0.88 | -0.29 | 0.86 | 0.50 | 0.50 | 0.20 | 0.91 | 0.58 | 1 | | | | | | | | | | |
| Cr | 0.46 | 0.70 | -0.72 | 0.72 | 0.68 | 0.90 | 0.00 | 0.41 | 0.62 | 0.26 | -0.36 | 0.89 | 0.60 | 0.72 | 1 | | | | | | | | | |
| Rb | -0.24 | 0.44 | 0.12 | 0.48 | 0.32 | 0.11 | 0.02 | 0.76 | 0.17 | 0.84 | 0.92 | 0.10 | -0.17 | 0.37 | -0.10 | 1 | | | | | | | | |
| Sr | 0.41 | 0.96 | -0.69 | 0.85 | 0.67 | 0.86 | -0.15 | 0.85 | 0.57 | 0.53 | 0.15 | 0.87 | 0.43 | 0.91 | 0.74 | 0.90 | 1 | | | | | | | |
| Y | 0.54 | 0.91 | -0.80 | 0.88 | 0.73 | 0.99 | -0.14 | 0.69 | 0.69 | 0.36 | -0.09 | 1.00 | 0.52 | 0.93 | 0.88 | 0.82 | 0.96 | 1 | | | | | | |
| Zr | 0.63 | 0.82 | -0.84 | 0.73 | 0.58 | 0.94 | -0.25 | 0.54 | 0.60 | 0.09 | -0.29 | 0.97 | 0.56 | 0.85 | 0.87 | 0.89 | 0.67 | 0.53 | 1 | | | | | |
| Ba | 0.03 | 0.90 | -0.32 | 0.71 | 0.46 | 0.59 | -0.27 | 0.97 | 0.28 | 0.70 | 0.54 | 0.62 | 0.38 | 0.84 | 0.43 | 0.86 | 0.97 | 0.99 | 0.59 | 1 | | | | |
| Hf | 0.58 | 0.85 | -0.82 | 0.75 | 0.59 | 0.95 | -0.26 | 0.59 | 0.58 | 0.16 | -0.26 | 0.98 | 0.61 | 0.87 | 0.90 | 0.14 | -0.13 | -0.33 | 0.53 | -0.29 | 1 | | | |
| Th | -0.55 | 0.20 | 0.48 | 0.17 | -0.01 | -0.20 | -0.11 | 0.58 | -0.16 | 0.69 | 0.99 | -0.19 | -0.19 | 0.15 | -0.35 | 0.16 | -0.20 | -0.31 | 0.26 | -0.26 | 0.40 | 1 | | |
| U | -0.08 | -0.03 | 0.10 | 0.07 | 0.18 | -0.18 | 0.49 | 0.11 | 0.12 | 0.47 | 0.31 | -0.24 | -0.45 | -0.19 | -0.10 | 0.07 | -0.16 | -0.35 | 0.48 | -0.32 | 0.99 | 0.28 | 1 | |
| Th/U | -0.64 | 0.16 | 0.56 | 0.11 | -0.08 | -0.24 | -0.18 | 0.55 | -0.25 | 0.65 | 0.98 | -0.21 | -0.10 | 0.14 | -0.37 | 0.93 | 0.95 | 0.85 | 0.83 | 0.87 | 0.17 | -0.08 | 0.14 | 1 |

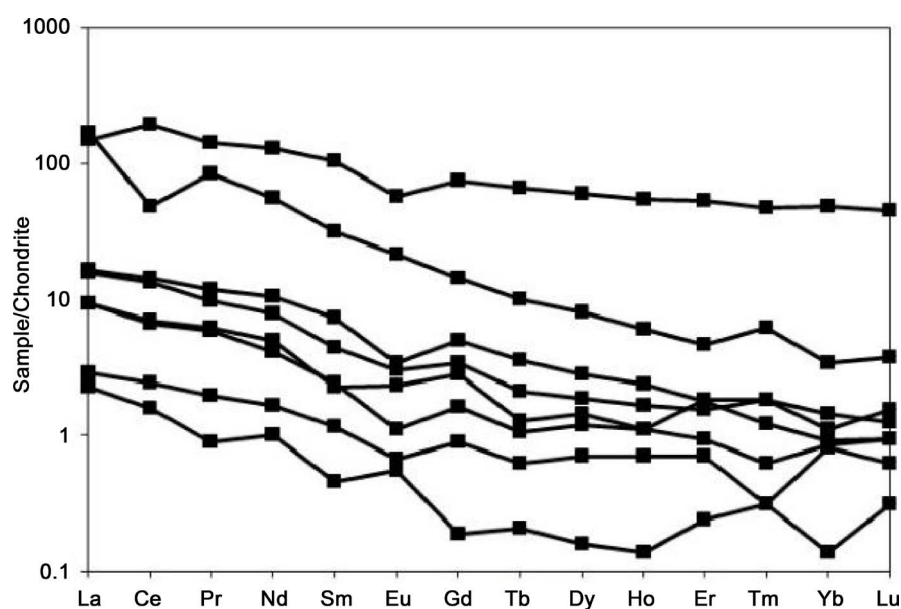


Figure 6. REE pattern of iron ore from the Edea north iron occurrence normalized with chondrite. Normalization values of chondrite after [28]. Samples are enriched in LREE compare depleted in HREE. There are also showing Eu anomalies that vary between negative and positive. Ce negative anomaly is conceding in one sample.

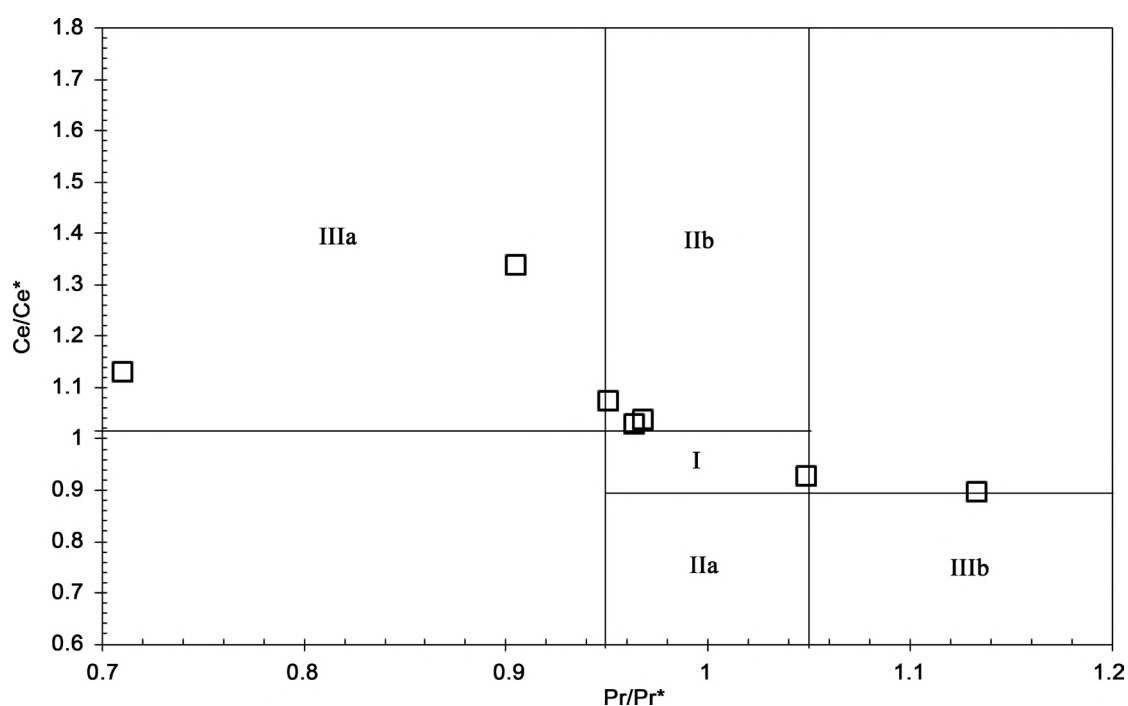


Figure 7. $(Ce/Ce^*)_{CN}$ VS $(Pr/Pr^*)_{CN}$ diagram (after [43]) for iron occurrence of the Edea North. Field I: neither Ce nor La anomaly; field IIa: positive La anomaly, no Ce anomaly; field IIb: negative La anomaly, no Ce anomaly; field IIIa: positive Ce anomaly; field IIIb: negative Ce anomaly.

Two main facies have been identified in the Edea North iron occurrence including the banded facies showing discontinuous banding and the massive facies. Based on the physical appearance and the chemical composition of the studied rocks, the name quartzite rich magnetite-martite has been proposed for this iron

formation.

The Precambrian iron formations have low Al_2O_3 , TiO_2 , P_2O_5 , CaO and MgO contents relative to post-Precambrian deposits [30] [31]. On a Fe_2O_3 - CaO + MgO - SiO_2 [30] and Al_2O_3 - SiO_2 - Fe_2O_3 [31] ternary diagrams, studied materials indicates a Precambrian affinity (Figure 8).

5.2. Textural Characterization of Magnetite

In the studied iron formations, magnetite presents irregular and elongated minerals which can contain quartz inclusions of various shapes and sizes. In addition to this textural observation, magnetite is deformed and may derive from chemical precipitations thus indicating a primary crystallization of magnetite [32]. The second magnetite texture is in the form of blasts clustering, around pyroxenes. It suggests a secondary crystallization resulting from the probable transformation of pyroxenes. Finally, xenomorphic magnetite crystals isolated in the silicate phases, which suggest the precipitation of magnetite during the formation of this occurrence. On the other hand, the relationships between magnetite and martite suggest the transformation of the first into the second during supergene enrichment. Overall, the following mineral transformation is suggested here: pyroxene-magnetite-martite.

5.3. Contribution of Detrital Materials

The participation of the detrital materials during the formation of iron mineralization has been highlighted in the literature. In the BIF, the contribution of detrital materials is generally indicated by the high concentrations of Al_2O_3 , TiO_2 and High Field Strength Elements (HFSE) (Zr, Hf, Ta, Th ...) and also by the inter-elements correlations between HFSE and rare earth ratios [10] [33]. The geochemical data of the Edea North iron occurrence are summarized in Table 2. High SiO_2 and Fe_2O_3 contents indicate chemical precipitation. In addition, [34] estimates that a typical chemical sediment is enriched in Mn and Fe, but the addition of detrital or volcanic materials results in dilution and enrichment of Ti, Al and Zr. However, the iron occurrence of the Edea North reveal relatively average values of Ti, Al and Zr which might suggest a low contribution of detrital materials. The correlation coefficients between Al_2O_3 and HFSE, LILE and some

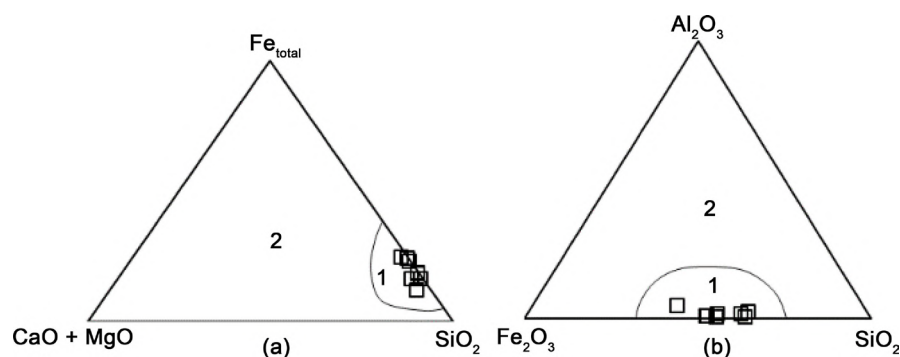


Figure 8. The North of Edéa iron occurrence composition. (a) After [31]. (b) After [30]. 1: Precambrian; 2: Post Precambrian.

transition metals Zr (0.82), Hf (0.85), Rb (0.44), Sr (0.96), Cr (0.70) and V (0.99) could also indicate a contribution of clastic materials during the deposition of these formations. The Th/U ratios in these samples range from 1.27 to 6.28, except for EDN17 where it is higher (29.25). This observation suggests a strong or moderate contamination by phosphates.

5.4. Origin and Enrichment Processes of the Edea North Iron Occurrence

Various methods have been used to establish the difference between hydrothermal, biogenic, detrital and seawater sources based on mineralogical, chemical and geochronological differences in the formation of iron deposits [6] [29] [30] [35] [36]. Among these methods, numerous discrimination diagrams have been proposed by several authors to highlight the influence of detrital components, hydrothermalism, biological components and seawater in their genesis [34] [37] [38] [39]. On the SiO_2 vs Al_2O_3 diagram [34], the iron mineralization samples from the Edea North iron occurrence plot within the hydrothermal domain (Figure 9). In the Fe-Mn-Al ternary diagram [38] the hydrothermal in origin is confirmed (Figure 10). The studied samples plot in the same area as those from the Bikoula BIFs of the Ntem complex [36]. This observation could indicate that Si and Fe of the studied deposits derived directly from a hydrothermal source. Similarly, in the Fe/Ti vs $\text{Al}/(\text{Al} + \text{Fe} + \text{Mn})$ diagram [37] a high affinity of the samples from the studied site with the hydrothermal domain is highlighted (Figure 11). This suggests an important contribution (in the volume) of the components of hydrothermal origin during the deposition of studied formations. Samples plot close to the field of modern metalliferous sediments indicating that some of the source components of the studied iron occurrence may have derived from sediments rich in highly hydrothermalized metals.

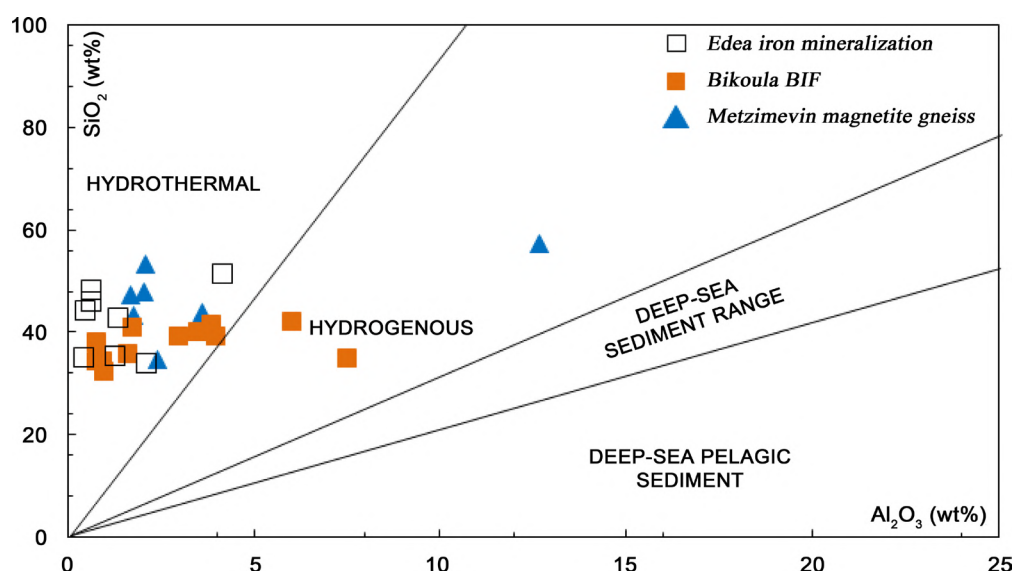


Figure 9. SiO_2 vs Al_2O_3 discrimination diagram indicating the hydrothermal affinity of the Edea iron mineralization. Note the hydrothermal affinity of Edea north samples; Metzimevin magnetite gneisses after [7] and Bikoula BIF after [36].

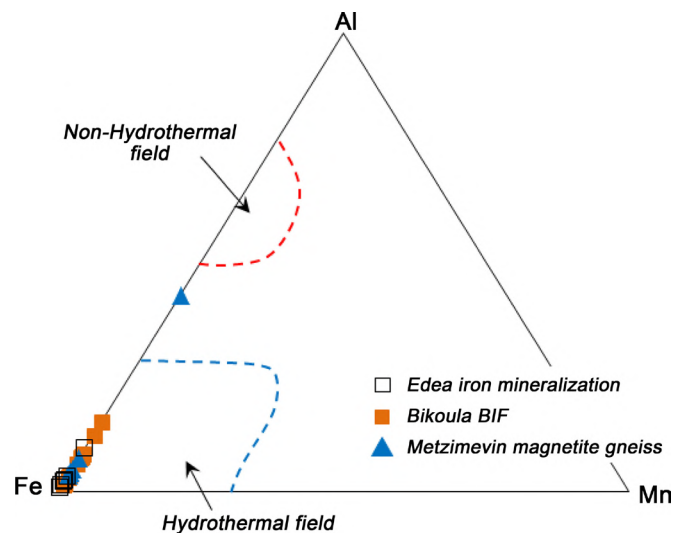


Figure 10. Ternary Fe-Mn-Al diagram indicating the hydrothermal affinity of the Edea North iron mineralization, Metzimevin magnetite gneisses (after [7]) and Bikoula BIF southern Cameroon (after [31]).

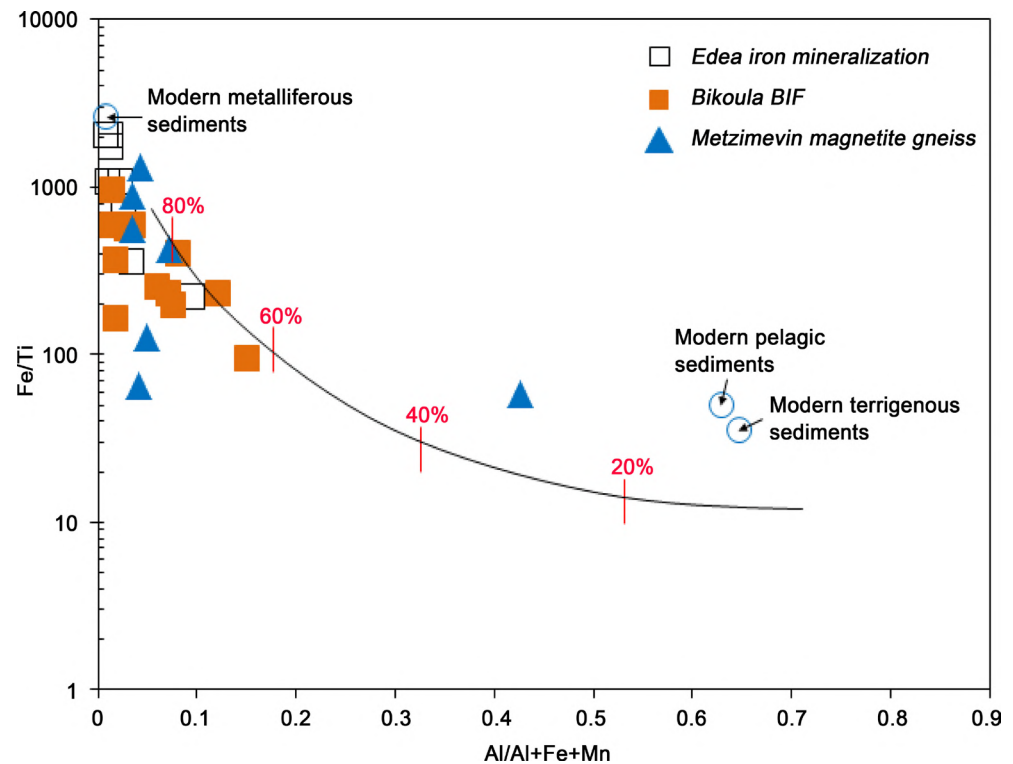


Figure 11. Composition of the Edea north iron occurrence plotted on the Fe/Ti vs Al/(Al + Fe + Mn) discrimination diagram after [37]. This diagram is proposed to estimate the relative involve of hydrothermal inputs in the fluids during the BIF precipitation.

The hydrothermal predominance is demonstrated particularly in the Y/P_2O_5 vs Zr/Cr discrimination diagram after [39] in which all samples have a low Zr/Cr and Y/P_2O_5 ratio (Figure 12). The vanadium content in the sediments is generally low when compared to the upper lithosphere [40]. The vanadium content varies between 5 and 69 ppm with an average of 20.5 ppm in the studied rocks

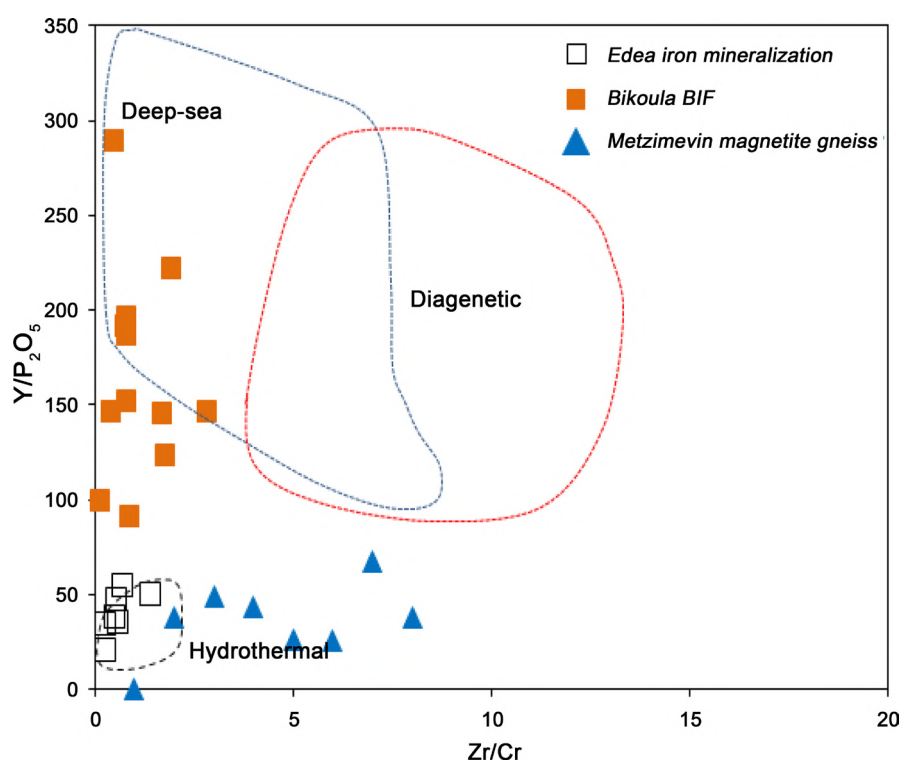


Figure 12. Composition of the Edea north iron mineralization plotted in a bivariate diagram of Y/P_2O_5 vs Zr/Cr . Deep Sea, diagenetic and metalliferous fields are based on data from [39].

(Table 2). This content is similar to that of deposits in China [41] and higher than BIF from the Ntem complex with values comprise between 9 and 11 ppm [42]. The Clarke value of chromium in sediments is 110 ppm and in igneous rocks from 200 to 2000 ppm. The chromium contents in the studied materials is smaller (10 and 80 ppm; average of 30 ppm) than the Clarke value of Cr in the sediments. Barium, on the other hand, shows concentration values between 1.6 and 131 ppm with an average of 51.17 ppm; this is relatively low compared to that of rocks of igneous origin. This content is comparable to sedimentary magnetite quartzites in the Arcot district of India [40]; thus a sedimentary origin can be inferred for the Edea North iron mineralization.

REE are commonly used by several authors to infer the environmental conditions for deposition and origins of iron ore formations [43] [44] [45] [46]. The REE content of the studied formation are relatively higher than REE content from iron formations in Cameroon (see [11] [36]). Chondrite-normalized REE patterns exhibit slight negative anomalies in Eu suggesting a prevalence of an oxidizing depositional environment during the Archean period [47]. The influence of the heterogeneous imprint of the clastic materials in the deposition basin could explain the negative Eu anomalies presented by most of the samples from this occurrence [48] [36]. Positive Eu anomalies represented by Eu/Eu^* with values between 1 and 1.83 demonstrates the contribution of deep-sea hydrothermal solutions [6] [49] [50]. Some of the samples show negative Ce anomalies (Figure 7). These negative anomalies suggest the influence of sea water into the system [51].

6. Conclusion

The origin and enrichment process of iron mineralization in Edea North are constrained by the coupling of petrographic and geochemistry analyses. The results obtained show that magnetite and quartz are the common predominant minerals while pyroxene and amphibole are secondary minerals. Based on this mineralogical composition, the studied rocks have been referred to as magnetite quartzite. In this occurrence, magnetite occurs in two textural varieties including irregular and elongated mineral blasts and finally xenomorphic in the silicate phase. The dominating chemical components of the studied materials are. A sedimentary origin with a high contribution of hydrothermal fluids and low input of detrital materials has been suggested for the Edea North iron occurrence.

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