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## Comparison of in-bottle measurements using <sup>15</sup>N and <sup>14</sup>C

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Field data reported in the literature are used for making a direct comparison between primary production estimated by the <sup>14</sup>C method and nitrogen uptake measured by the <sup>15</sup>N method (nitrate, ammonium, and, at times, urea). C/N uptake ratio values vary widely, from less than 1 to more than 20 (at/at), without any evident relationship with environmental properties or experimental procedure. The mean uptake ratio calculated on the whole data set is found to be equal to 6.7 (n = 99). The question arises whether such a C/N ratio may be considered as the assimilation ratio. For comparison purposes, the constituent ratio of phytoplankton assessed by the slopes of the regression lines relating carbon and nitrogen in particles has also been studied from the literature. This C/N ratio ranges between 5 and 7.5 (at/at) in the ocean, whatever the region, the season, or the phytoplankton present, suggesting that in the field the phytoplankton constituent ratio is probably never very different from the mean ratio found in phytoplankton by Redfield. Different possible sources of uncertainty are discussed to explain discrepancies between C/N uptake values and composition ratios. A lack of standardization in analytical procedures and calculations is evident. A better knowledge of the accurate value of the C/N assimilation ratio in the different oceanic systems is needed, especially for reaching the new production ratio "f" by using the <sup>15</sup>N and <sup>14</sup>C methods.

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## Introduction

Since recognition of the potential effect of human activities on the natural evolution of climate, one of the main objectives of most large-scale aquatic ecological research programmes has been to understand the main properties of the ocean and to quantify the global flux of carbon due to biological processes. Undoubtedly, satellites are the most appropriate tools for providing maps of surface chlorophyll distribution derived from ocean colour (Smith and Baker, 1982; Platt and Herman, 1983; Feldman et al., 1984). Algorithms have been proposed for estimating primary production from these satellite images (Smith et al., 1982; Eppley et al., 1985; Platt, 1986; Platt and Sathyendranath, 1988). Furthermore, new production, which represents the exportable component of total production (Dugdale and Goering, 1967), can also be estimated from remote sensing, at least in certain propitious conditions (Dugdale et al., 1989; Sathyendranath et al., 1991). Basically, all the algorithms used to calculate both total and new production clearly require for their construction a fair knowledge of the main primary production processes, and as accurate a "sea truth" as possible. However, despite decades of field studies, factors controlling phytoplankton growth and production are not yet well defined in all aquatic systems, and the absolute magnitude of total and new production is still often wrapped in uncertainty, especially in the case of oligotrophic waters. This is because of the difficulty, both conceptual and technical, of evaluating the daily production of phytoplankton in the natural environment (Peterson, 1980).

Among the numerous methods developed for measuring the primary production of phytoplankton in terms of carbon and nitrogen, use of the tracers <sup>14</sup>C and <sup>15</sup>N has largely predominated since their introduction by Steemann Nielsen (1952) and Dugdale and Goering (1967) respectively. The <sup>14</sup>C method is the standard technique used for estimating primary production in nearly all aquatic ecology laboratories. Relatively easy, rapid, sensitive and precise, this method has received a

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number of modifications, reducing many of its original imperfections and increasing the accuracy of the results. One of the recent prevailing recommendations has been the "clean technique" (Fitzwater *et al.*, 1982).

The <sup>15</sup>N method is much more delicate, tedious, and time-consuming and requires accurate assessments of dissolved inorganic and particulate organic nitrogen. However, it has been extensively used in many laboratories. Dugdale and Wilkerson (1986) have reviewed most of the difficulties relevant to the method, and have proposed some recommendations for obtaining reliable estimates of phytoplankton nitrogen flux in eutrophic areas. In spite of its difficulties, the <sup>15</sup>N method is the only one allowing the distinction between new production based on nitrate uptake from regenerated production based for the most part on ammonium utilization (Dugdale and Goering, 1967).

When primary production  $({}^{14}C)$  and new production  $({}^{15}N)$  have been measured, the percent new production (Dugdale and Goering, 1967), of "f" factor such as defined by Eppley and Peterson (1979), can be calculated with the relationship:

$$f = \rho^{15} N - NO_3 \times 6.6 / \rho^{14} C$$

 $\rho$  is the rate of C or N uptake, and 6.6 is the assimilation ratio C/N (at/at) assumed equal to the constituent ratio determined in phytoplankton by Redfield (1958).

If "f" and primary production are known, "f" provides direct access to the rate of new production, which is the true fertility index of an area. Primary production is expressed in terms of carbon while new production is measured in terms of nitrogen. Hence, the relationship between carbon fixation and nitrogen uptake has to be clearly defined in the field in order to calculate "f".

When all the forms of N uptake are assessed simultaneously, the rate of total nitrogenous production is obtained. Since the <sup>14</sup>C method gives the rate of total carbon fixation, the absolute magnitude of primary production can be approached by two different ways, the <sup>15</sup>N and <sup>14</sup>C methods. A direct comparison between the two is then possible.

## The C/N composition ratio

Comparison between <sup>14</sup>C and <sup>15</sup>N utilization requires knowledge of the range of values in which the C/N assimilation ratio is expected to vary. If the physiological processes leading to the synthesis of new cellular material were in balance during the course of experiments, then assimilation ratios would resemble those of cellular composition (McCarthy and Nevins, 1986a). In cultures of phytoplankton under nutrient or light limitation, the C/N constituent ratio may vary within brief periods from the Redfield ratio of 6.6 to about 20, or more, according to the growth rate limitation (Goldman, 1984). In the field, the C/N constituent ratio of phytoplankton is not always directly measurable, because of the difficulty of separating algal cells from the non-algal particulate material. However, the relative concentrations of particulate carbon (PC) and particulate nitrogen (PN) in the field are considered to reflect \* the chemical composition of phytoplankton (Goldman *et al.*, 1979).

The regression lines linking (PC) and (PN) calculated for a great number of different oceanographic situations are reported in Table 1. The residual carbon for N = 0would correspond to carbonaceous compounds that do not covary with nitrogen (Copin-Montegut and Copin-Montegut, 1983). The slopes of the regression lines represent the C/N ratio due to phytoplankton, but also in part to zooplankton and some microheterotrophs and detritus. However, it is assumed that these slopes give a good indication of the C/N constituent ratio of phytoplankton, especially when phytoplankton are known to predominate, as for example during coastal upwelling blooms. From the data in Table 1 it appears that differences between regions and systems are very small, even when coastal upwelling areas are compared with the extreme oligotrophic areas, such as the tropical Atlantic or the southwestern Pacific between 7°S and 15°S. The slopes of the regression lines C/N vary in a narrow range, from 5.1 to 7.6 (model II), and are generally close to the Redfield ratio. Interestingly, Herbland and Le Bouteiller (1983) and Eppley et al. (1992) have tested the effect of the nutrient abundance in stratified situations and found that both slopes and intercepts of the relationships between PC and PN were independent of the presence or absence of nitrate. Within the euphotic layer, PC/PN does not seem to exhibit any significant vertical gradient with respect to light or nutrients.

In the Guinea Dome, the size distribution of PC/PN in seston indicates that the  $<1 \,\mu$ m fraction shows the same properties as the total fraction (Table 1). Assuming that the non-algal matter has no particular effect on C/N, the chemical composition of picophytoplankton, which represents nearly 50% of total phytoplankton in that area (Le Bouteiller *et al.*, 1992), would not be different from large algal cells.

All these results provide clear evidence of a striking consistency in the chemical structure of the organic matter. This suggests that in the oceanic photic layer the constituent ratio of phytoplankton would be close to the C/N ratio of algal cells growing at a maximum growth rate (Goldman, 1980). From an empirical approach, there are many indications that, at least on average, the assimilation ratio should not be very different from the C/N Redfield ratio. Table 1. Linear regression analysis between carbon and nitrogen in particles (PC = Y; PN = X, both in micromoles). Slope and Yintercept were calculated with the model I method. When X and Y are simply measured and are not under the control of the investigator, model II is more appropriate (Sokal and Rohlf, 1969). The slope of the regression line using model II is also given. PIRAL and PROPPAC cruises were conducted by ORSTOM, France (unpublished data). Size fractionations during PIRAL cruises were presented by Fukaï (1991) and Le Bouteiller *et al.* (1992).

		Model I						
Area		Slope	Slope Y-int n r		r	Model II Slope	Reference	
<sup>h</sup> Indian Ocean 23	3°S to 46°S	5.45	0.59	82	0.996	5.47	Copin-Montegut and	
Antarctic Ocean	1	5.49	0.10	62	0.996	5.51	Copin-Montegut (1983)	
. Tropical Atlanti	ic	5.21	0.63	29	0.977	5.33	Copin-Montegut (1983)	
Morocco upwell	ling	5.32	2.55	60	0.971	5.48	Copin-Montegut (1983)	
Mauritania upw	elling	4.93	1.66	55	0.973	5.07	Copin-Montegut (1983)	
Mediterranean S	Sea	5.89	0.30	296	0.975	6.04	Copin-Montegut (1983)	
All data		5.44	0.73	658	0.980	5.55	Copin-Montegut (1983)	
Equat. Atlantic	0°,4°W	6.00	0.75	100	0.92	6.52	Herbland and	
NO <sub>3</sub> -rich water	0°,4°W	6.08	0.64	71	0.89	6.83	Le Bouteiller (1983)	
W. Pacific 15°S	173°E	5.26	1.29	48	0.81	6.49	Lemasson, Cremoux, 1985	
Guinea Dome 1	2°N 22°W	6.18	1.34	107	0.92	6.72	PIRAL cruises, 1986	
$<1 \mu m$ , 12°N 22	°W	5.23	1.44	97	0.77	6.79	PIRAL cruises, 1986	
Pacific 5°N-14°S	5165°E	6.12	0.85	137	0.81	7.56	PROPPAC cruise, 1987	
Upwelling, Paci	fic 0°							
165°E		5.65	1.48	91	0.79	7.15	PROPPAC cruise, 1988	
Pacific 7°S 165°H	Ξ	5.15	1.10	81	0.75	6.87	PROPPAC cruise, 1989	
Pacific 15°N-12°	°S 135°W	5.70	1.60	67	0.82	6.95	Peña et al. (1991)	
Pacific 15°N–15°	°S 150°W	4.98	1.21	64	0.81	6.15	Eppley et al. (1992)	

## The C/N assimilation ratio

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Still today it is not easy to carry out and obtain reliable measurements of production with either the <sup>14</sup>C method (Gieskes and Kraav, 1984; Grande et al., 1989) or the <sup>15</sup>N method (Glibert et al., 1982; Harrison, 1983; Laws et al., 1985: Garside, 1991). The variance within a data set is often fairly small (Le Bouteiller and Herbland, 1984; Lohrenz et al., 1988b; Williams and Purdie, 1991), but comparisons between different data sets are still disappointing (Richardson, 1991). Moreover, in most oceanic systems large changes in primary production rate are apt to occur within the course of a few days (Herbland and Le Bouteiller, 1982; Lohrenz et al., 1988a; Platt et al., 1989; Fukai, 1991), such that comparison between C and N uptake measured at different times may have no real significance. Finally, the sine qua non condition to achieve for a reasonable comparison between C and N production consists in considering only the data sets of  $\rho^{14}$ C and  $\rho^{15}$ N which have been obtained simultaneously in well-controlled experimental conditions.

Most research reported in the literature deals with total nitrogen production of phytoplankton in oceanic waters, generally <sup>15</sup>N-NO<sub>3</sub> and <sup>15</sup>N-NH<sub>4</sub> uptake, and, at times, <sup>15</sup>N-urea uptake. Some research also includes simultaneous <sup>14</sup>C fixation measurements. A strict comparison between C and N production is then possible.

The data summarized in Table 2 concern a great number of regions, different seasons, and various situations. The range of variations in  $\rho^{14}C/\rho^{15}N$  (Table 2)

appears clearly much larger than that of PC/PN (Table 1).

There are two explanations for this discrepancy: (1) The uptake ratio C/N may change according to the environmental factors, thus reflecting the physiological response of the natural assemblages of phytoplankton to the experimental conditions during incubation. (2) When used in the field, the <sup>14</sup>C or <sup>15</sup>N methods over- or underestimate the actual rate of C or N production, for many different technical reasons; the main ones will be detailed later.

Some of the first values in Table 2 (for example MacIsaac and Dugdale, 1972; McCarthy, 1972; Mac-Isaac et al., 1974) reflect a large predominance of C fixation over N uptake, whatever the studied system. When  $\rho^{14}C/\rho^{15}N$  exceeds 15 or 20 (atomic), experimental procedures are probably to be suspected rather than a lack of balance between uptake and composition. Paradoxically, several studies in nutrient-rich waters of coastal upwellings present also relatively high  $\rho C/\rho N$ : Morocco (Slawyk et al., 1978), Mauritania (Slawyk et al., 1978), Peru in 1976 (Wilkerson et al., 1987), and Namibia (Probyn, 1988, and Estrada and Marrase, 1987). It is difficult to suggest that nitrogen deficiency might explain these observations.

Conversely, the data summarized in Table 2 suggest that nutrient-depleted waters do not involve significantly higher C/N uptake ratios in the open ocean. This is to be compared to the lack of clear effect of oligotrophic conditions upon the PC/PN ratio (Table 1).

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Table 2. Assimilation ratios C/N (at/at) calculated from primary production ( $^{14}$ C method) measured in sea water in parallel with total nitrogen production ( $^{15}$ N method): nitrate, ammonium and, at times, urea uptake. IS and SIS: *in situ* and "simulated" *in situ* incubations.

					C/N	
Position	Reference	<sup>15</sup> N-compounds	Incubation	Mean	Range	<u>n</u>
Atlantic 26°-42°N 65°W	Dugdale and Goering (1967)	NO3,NH4	24 h SIS	9.5	1.6–34	7
Upwelling Peru	MacIsaac and Dugdale (1972)	NO <sub>3</sub> NH <sub>4</sub>	24 h SIS	11.7	6.4-22	6
Eastern Pacific	MacIsaac and Dugdale (1972)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	75.7	14–178	4
Eastern Pacific	MacIsaac and Dugdale (1972)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	27.7	_	1
Eastern Mediterranean	MacIsaac and Dugdale (1972)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	20.8	-	1
Western Mediterranean	MacIsaac and Dugdale (1972)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	15.7	7.8–46	8
Coast of California	McCarthy (1972)	NO <sub>3</sub> ,NH <sub>4</sub> , urea	24 h SIS	11.2	8.8-14.6	4
N. Pacific Central Gyre	Eppley et al. (1973)	NO <sub>3</sub> ,NH <sub>4</sub> , urea	24 h SIS	5.07	3.0-7.65	3
Upwelling NW Africa	MacIsaac et al. (1974)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	15.5	11.7–24	4
Upwelling Morocco	Slawyk et al. (1978)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	13.7	4.1-31.2	5
Upwelling Mauritania	Slawyk et al. (1978)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	32.6	26.3-36.9	4
Saanich Inlet	Harrison (1978)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	4.92		1
Southern California Bight	Harrison (1978)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	6.88	-	1
Eastern Canadian Arctic	Harrison et al. (1982)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h SIS	4.74	1.8-7.4	14
Middle Atlantic Bight	Harrison <i>et al.</i> (1983)	NO <sub>3</sub> ,NH <sub>4</sub>	6 h SIS	4.06	2.6-7.5	8
	Falkowski <i>et al.</i> $(1983)$	NO NU	04 L 670	0.07	2 4 26 7	7
Eastern Canadian Arctic	Harrison <i>et al.</i> (1985)	$NO_3, NH_4, urea$	24 n 515	8.27	3.4-20.7	1
NW Atlantic 40°N 65°W	McCarthy and Nevins (1986a)	$NO_3, NH_4$ , urea	3-4 n SIS	9.30	4.3-23	ິ
NW Atlantic 40°N 65°W	McCarthy and Nevins (1986b)	$NO_3, NH_4$ , urea	4 n SIS	7.60	5.1-9.5	2
Upwelling Peru, 1976	Wilkerson et al. (1987)	$NO_3, NH_4$	6 h SIS	14.9		29
Upwelling Peru, 1977		$NO_3, NH_4$	6 h SIS	7.07		14
NW Atlantic, coast mixed	Harrison and Wood (1988)	$NO_3, NH_4$	24 h SIS	5.56		1
NW Atlantic, frontal	~	$NO_3, NH_4$	24 h SIS	7.65	5.83-9.48	2
NW Atlantic, stratified	Harrison and Wood (1988)	$NO_3, NH_4$	24 h SIS	11.5	1.93-21	2
NW Atlantic, oceanic	Harrison and Wood (1988)	$NO_3, NH_4$	24 h SIS	10.1	6.60 - 12.8	4
Upwelling Namibia	Probyn (1988), Estrada and Marrase (1987)	NO <sub>3</sub> ,NH <sub>4</sub> , urea	4-6 h SIS	12.1	_	10
Eastern equatorial Pacific	Murray et al. (1989)	NO <sub>2</sub> NH	24 h IS	7 12	37-112	4
Antarctic spring 1983	Smith and Nelson (1990)	NO <sub>2</sub> NH	24 h SIS	4 17	2 2-6 4	15
Antarctic autumn 1986	Shifth and Heison (1990)	NO <sub>3</sub> , NH	24 h SIS	2 43	0.75-5.0	6
Eastern Pacific VERTEX	Knauer $et al$ (1990)	NO <sub>3</sub> ,NH <sub>4</sub>	24 h 010	5 25	1.96-10.2	6
Pacific 8°N_15°N 150°W	Dugdale <i>et al.</i> $(1990)$	NO <sub>2</sub> NH.	6 h SIS	6.80	1.50 10.2	4
Pacific 1°N 6°N 150°W	Dugdale <i>et al.</i> $(1992)$	NO <sub>3</sub> ,NH <sub>4</sub>	6 h SIS	7 50		4
Pacific 0º 150°W	Dugdale <i>et al.</i> $(1992)$	NO. NH.	6 h SIS	7.50	_	4
Pacific 2°S 8°S 150°W	Dugdale <i>et al.</i> $(1992)$	NO. NH.	6 h SIS	4 10		4
Pacific 10°S–15°S 150°W	Dugdale et al. $(1992)$	$NO_3, NH_4$	6 h SIS	0.94		3

 $\rho C/\rho N$  were often observed below 5 (atomic) in the eastern Canadian Arctic (Harrison *et al.*, 1982, 1985) and also in the marginal ice zone of the Weddell Sea (Smith and Nelson, 1990). This leads to the hypothesis that the C/N uptake ratio would be specifically lower than the Redfield ratio in subpolar waters. This result is not corroborated by PC/PN analyses in the Antarctic Ocean (Table 1), but low  $\rho^{13}C/\rho^{15}N$  uptake ratios were also observed at 60°S and 66°S by Collos and Slawyk (1986).

When all the data from Table 2 are considered (except those from Dugdale and Goering, 1967; MacIsaac and Dugdale, 1972; Slawyk *et al.*, 1978), a good correlation between  $\rho^{14}$ C and  $\rho^{15}$ N (total) is obtained (r = 0.87; n = 104). The mean assimilation ratio is 6.7 (C/N = 6.70 ±

0.97; n = 104). One may conclude that, on the average, the C/N assimilation ratio is not significantly different from the expected Redfield ratio. This is undoubtedly an important result.

# Sources of uncertainty in C/N uptake ratio

The results in Table 2 have not all been obtained by exactly the same methodology. Some experiments were carried out *in situ* while most of them were "simulated" *in situ* incubations, which could have some consequences (Lohrenz *et al.*, 1992), although not necessarily appreciable, upon the uptake ratio C/N (Slawyk *et al.*, 1976). The length of incubation time is probably the

most important factor to consider for the present comparison. Short-term and long-term experiments lead to an approximation of gross and net production respectively. Many studies in the field have shown that C and N have not the same evolution within bottles during incubations, from the uptake by photosynthetic and possibly heterotrophic processes, till incorporation into autotrophic and, depending upon the ecosystem, into heterotrophic organisms. Brief uptake experiments seem to lead to a predominance of carbon uptake (Slawyk *et al.*,

1978). Conversely, in nitrogen limited waters, the ammonium incorporation by phytoplankton is potentially very rapid (Glibert and Goldman, 1981; Dortch et al., 1982). Hence, low short-term C/N uptake ratios may occur. During daytime, photosynthesis directly depends on available light, whereas nitrogen uptake may occur at night. In the open ocean, NO<sub>3</sub> is almost never taken up in darkness (Nelson and Conway, 1979), except in very nitrogen-poor waters (Cochlan et al., 1991), while dark NH<sub>4</sub> uptake seems much more important, varying from 10 to 100% of light uptake (Cochlan et al., 1991). However, a significant amount of inorganic N uptake is probably due to heterotrophic bacteria (Laws et al., 1985; Harrison et al., 1992). Because of nocturnal respiration and excretion of carbon (Eppley and Sharp, 1975; Laws et al., 1987; Harris et al., 1989), it appears evident that processes occurring at night are very important to take into account here. As a consequence, a 24 h incubation seems to give a better representation of average daily rates of carbon and nitrogen assimilation (Eppley et al., 1973).

Nevertheless, in addition to the light effect, several problems arise when the incubation time exceeds about 12 h, mainly related to respiration, excretion, and regenerative processes, such that the C and N uptake rates may become non-parallel. When long incubations are performed, the risk of a "bottle effect" increases (Gieskes et al., 1979; Chavez and Barber, 1987; Wilkerson and Dugdale, 1992), especially when very delicate picophytoplankton are present (Murphy and Haugen, 1985). Besides, the <sup>15</sup>N uptake rate may increase during incubation due to <sup>15</sup>N addition in the sample, which needs to be corrected by the use of the Michaelis-Menten equation (Paasche and Kristiansen, 1982). Ammonium regenerated during incubation may also influence N flux measurements by causing underestimation of the uptake rate because of isotope-dilution of the <sup>15</sup>NH<sub>4</sub> concentration (Glibert *et al.*, 1982). Laws (1984) has shown that this isotope dilution was overestimated by Glibert et al. (1982) owing to an inappropriate method of calculation, and because of using Reeve Angel filters through which many picoplankton pass easily (Le Bouteiller et al., 1992). Harrison and Harris (1986) also found that isotope dilution was significant and, if ignored, could account for 1.5 to 3-fold underestimates in computed uptake rates. Kanda *et al.* (1987) described a new practical calculation procedure to correct the underestimate caused by isotope dilution.

Moreover, a discrepancy has been observed between disappearance of nitrate or ammonium such as measured chemically, and uptake of the same compounds by the <sup>15</sup>N method (Price *et al.*, 1985; Dugdale and Wilkerson, 1986; Slawyk *et al.*, 1990). Although not fully explained, the difference between these two approaches leads to a limit on incubation periods of 6 h or less. Many investigators, however, still use 12 or 24 h.

The consequence of relatively short experiments, where incubation artifacts are limited, is that an extrapolation is needed to obtain daily production. Experiments summarized in Table 2 comprise incubation periods ranging from 3 to 24 h. Many results of short incubations have been simply expressed as mM  $m^{-3} h^{-1}$  (Harrison *et* al., 1983; Wilkerson et al., 1987; Harrison and Wood, 1988; Probyn, 1988), which is not fully satisfactory for the present purpose. McCarthy and Nevins (1986a) converted the hourly rates measured during a 3 or 4 h incubation period to a daily rate by multiplying both nitrate and ammonium uptake rates by 24. Dugdale et al. (1992) chose another procedure: after 6 h incubation, the hourly uptake rates of NO3 and NH4 were multiplied by 12 and 18 respectively to obtain a daily rate. Knauer et al. (1990) observed that after 6, 12, and 24 h time-course experiments NO<sub>3</sub> uptake did not deviate from linearity. However, NH<sub>4</sub> uptake rates over 1-6 h were approximately 2.5 times greater than those after 24 h. Based on these findings, Knauer et al. (1990) have not corrected NO3 uptake rates after 24 h incubations, but NH4 uptake rates have been revised upwards and multiplied by a factor of 2.5 to correct for the observed non-linearity.

All these various analytical procedures and calculations may explain part of the discrepancy between carbon fixation and total nitrogen uptake (Table 2). Great differences appear clearly in the corrections applied to the calculation of daily rates, taking into account both isotope dilution and length of incubation. A lack of standardization in methods is evident.

Several authors have measured urea taken up by phytoplankton in addition to nitrate and ammonium (Table 2). In the nitrate-rich coastal waters, urea represents 10 to 30% of total N uptake (McCarthy *et al.*, 1977; Probyn, 1985; L'Helguen, 1991), but may become negligible when nitrate concentration is high (Kokkinakis and Wheeler, 1987, 1988). In Arctic and Antarctic waters, the urea uptake rate ranges from 10 to 50% (Probyn and Painting, 1985; Wheeler and Kokkinakis, 1990). Phytoplankton have probably the highest affinity for urea in oligotrophic regions (Eppley *et al.*, 1977; Kanda *et al.*, 1985; Price and Harrison, 1988). For example, the urea contribution in the Central North Pacific Gyre ranges from 27 to 57% of total N uptake,

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generally more than nitrate uptake (Eppley *et al.*, 1973; Sahlsten, 1987). These results provide clear evidence that in a number of regions, and especially in oceanic areas, urea utilization by phytoplankton represents at least 10% and usually 20 or 30% of total N uptake, which is far from negligible when all the forms of dissolved nitrogen used by phytoplankton are taken into account. As mentioned above for other factors, the results in Table 2 do not show significantly lower  $\rho^{14}C/\rho^{15}N$  when urea uptake has been measured in addition to nitrate and ammonium. Too many causes of variations probably interact in a relatively too small data set.

Dissolved organic nitrogen also contributes to the nutrition of phytoplankton (Antia *et al.*, 1991). However, recent studies have shown that in the oligotrophic northeast Pacific, dissolved organic nitrogen flux represented a relatively insignificant component of the nitrogen balance in the surface ocean (Harrison *et al.*, 1992).

### The data quality

Actually, most sets of  $\rho C/\rho N$  ratios in Table 2 present wide variations around the Redfield ratio, and so any systematic over- or underestimation of C or N production rate is impossible to detect. The number of chemical and biological parameters to be measured for only one comparison is undoubtedly the source of much uncertainty, especially in oligotrophic systems where most of the methods reach their analytical limits. The diversity of experimental procedures used by authors (see above) is another evident cause of discrepancy.

In oceanic primary productivity studies there is no suitable diagnostic index by which the production of different areas could be compared and tested. There is no means for deciding if, in one given time and space, the true rate of primary production such as measured by the <sup>14</sup>C or <sup>15</sup>N technique has actually been achieved. Even the amount of production data collected in a region is not the required criterion of quality because of possible systematic bias.

However, there are two approaches often used to test the level of *in situ* production. The first consists in standardizing production by light (Platt, 1969), chlorophyll (Platt and Subba Rao, 1973), or by both light and chlorophyll (Morel, 1978; Platt *et al.*, 1988). The other approach requires the results of different methods run simultaneously to be compared ( ${}^{14}C$ ,  ${}^{13}C$ ,  ${}^{15}N$ ,  $O_2$ , . . .).

Production has generally been divided by chlorophyll in the first approach. In the tropical open ocean, and probably also in many other ecosystems, the vertical profiles of productivity index (g C g<sup>-1</sup> Chl a h<sup>-1</sup>) seem always to present the same typical pattern, whatever the ICES mar. Sci. Symp., 197 (1993)

nitrate depth distribution may be (Le Bouteiller and Herbland, 1984): a low value at the surface due to inhibition of photosynthesis by light excess, a subsurface maximum at 10 or 20 m depth (20–30% of surface light), and a very regular decrease downwards, down to the bottom of the photic zone. Tests of data quality and regional comparisons are possible with the productivity index, and especially with the maximum value in the profile.

For example, Thomas (1970) observed that in nutrient-poor waters of the tropical Pacific Ocean, the productivity index was near 3 on average, and up to 5 g C  $g^{-1}$  Chl  $a h^{-1}$  in rich waters, but "the difference between the two was not great". More recently, much higher subsurface productivity indices have been measured in deep blue tropical seas: 7.3 to 18.9 g C  $g^{-1}$  Chl  $a h^{-1}$  (Williams *et al.*, 1983), 11.6 to 15.3 (Herbland and Le Bouteiller, 1983), up to 15 (Gieskes and Kraay, 1986), and around 10 g C  $g^{-1}$  Chl  $a h^{-1}$  (Laws *et al.*, 1984, 1987, 1990). A mean maximum productivity index of only 59 g C  $g^{-1}$  Chl  $a d^{-1}$  was observed by Cullen *et al.* (1992) in the equatorial Pacific at 155°W.

Comparison of primary production measurements using  $^{14}$ C and  $^{15}$ N methods (Table 2) has led to the conclusion that, on average, the assimilation ratio C/N is not significantly different from the composition ratio (Table 1). This approach, however, cannot be used to test the absolute magnitude of C or N production rate without the certainty that in the system studied the C/N assimilation ratio is close to the Redfield ratio. This statement clearly must be reinforced in the future.

## The f ratio

Total primary production  $({}^{14}C)$  may be compared indirectly with nitrogen production  $({}^{15}N)$  through the "f" factor (Eppley and Peterson, 1979), which can be obtained in three different ways:

$f = \rho^{15} N - NO_3 / \rho^{15} N - (NO_3, NH_4, urea)$	(1)
$f = \rho^{15} N - NO_3 \times 6.6 / \rho^{14} C$	(2)

$$f = \rho^{15} N - NO_3 \times 6.6 / \rho^{13} C$$
(3)

Provided primary production is consistently estimated by the methods using  $^{14}C$ ,  $^{13}C$ , or  $^{15}N$ , then "f" must be found equivalent in a given system whatever the tracer used.

Table 3 gives some recent values of "f" measured in the tropical and subtropical Atlantic and Pacific open oceans, where the seasonal signal is always expected to be relatively low. All results are remarkably consistent and independent of the methods used. Knauer *et al.* (1990) have compared "f" obtained with relationships 1 and 2, and found 0.11 and 0.16 respectively for a large 1CES mar. Sci. Symp., 197 (1993)

Position	Date	Reference	Incubation	Methods	f-ratio
13°N 22°W	Oct 1976	Slawyk and	SIS	<sup>13</sup> C <sup>15</sup> NO <sub>3</sub> <sup>15</sup> NH <sub>4</sub>	0.02-
500 · 0003		Collos (1982)		<sup>13</sup> NO <sub>2</sub> <sup>13</sup> N-urea	0.20
5°S to 30°N	Nov 1982	Kanda <i>et al.</i> $(1985)$	SIS	$^{13}C$ $^{13}NO_3$	0.11
170°E to 110°W	Feb 1983	Kanda <i>et al.</i> $(1985)$	10	140 1510	0.05
0'4 W	Feb 1979	Le Bouteiller (1980)	15	$M_{\rm NO_3}$	0.05-
200NI 1550W	Jan 1980	Le Boutellier (1980)	CIC	15100 15111	0.25
20 IN 155 W	Aug 1965	Sanisten (1987)	515	$^{15}N$ urea	0.14
2°305 84°W	Jun 1987	Murray et al (1080)	15	<sup>15</sup> NO. <sup>15</sup> NH	0.20
0°558 87°W	5un 1707	Murray <i>et al.</i> $(1989)$	10		0.22
2°47N 86°W		Murray <i>et al.</i> $(1989)$			0.24
4°26N 85°W		Murray <i>et al.</i> $(1989)$			0.17
32°N to 39°N	Jun Jul	Harrison (1990)	IS	<sup>15</sup> NO3 <sup>15</sup> NH4	0.07-
55°W to 65°W				5 +	0.28
9°N to 10°N	Mar Apr	Harrison (1990)	SIS	<sup>15</sup> NO <sub>3</sub> <sup>15</sup> NH <sub>4</sub>	0.43-
89°W to 94°W	-				0.84
33°N 139°W	1987–1988	Knauer et al. (1990)	IS	<sup>15</sup> NO <sub>3</sub> <sup>15</sup> NH <sub>4</sub>	0.11-
				<sup>14</sup> C	0.16
150°W	Feb Mar	Dugdale <i>et al</i> . (1992)	SIS	<sup>14</sup> C <sup>15</sup> NO <sub>3</sub>	
8°N to 15°N	1988				0.06
$1^{\circ}N$ to $6^{\circ}N$		Dugdale et al. $(1992)$			0.13
0°		Dugdale <i>et al.</i> $(1992)$			0.24
2-5 to 8-5		Dugdale <i>et al.</i> $(1992)$			0.13
10'5 10 15'5	A	Dugdale <i>et al.</i> $(1992)$	ore	1401510	0.07
155 W 19N to 169N	Apr 1988	Pena et al. (1992)	212	$-C^{-}NO_{3}$	0.22
2°N to 6°S		Penn et al. (1992)			0.22
12°S to 15°S		Penn et al. (1992)			0.39
		1 cha ci ul. (1772)			0.09

Table 3. Some recent results of f-ratio measurements in the tropical and subtropical Atlantic and Pacific Oceans. New production estimated by the <sup>15</sup>N method, and total production estimated by the <sup>15</sup>N, <sup>14</sup>C, or <sup>13</sup>C methods.

number of data from the Vertex time-series site. At 150°W, Dugdale et al. (1992) also found general consistent agreement between total production values computed from <sup>15</sup>N and <sup>14</sup>C measurements, except south of 10°S where "f" values calculated with Equation (2) were about four times higher than values calculated with Equation (1). This observation probably results from very low primary production values estimated with the <sup>14</sup>C method from 10°S to 15°S, as confirmed by other recent <sup>14</sup>C results (Barber and Chavez, 1991) in similar conditions. Such an underestimation of the rate of primary production (<sup>14</sup>C method) evidenced by comparison with the nitrogen production (<sup>15</sup>N method) measured simultaneously gives an example of common problems arising in primary productivity experiments. The interest in direct comparison between the results of two independent methods used in parallel is thus clearly demonstrated. Some experimental artifacts are difficult to put into evidence in the field studies. They are probably more frequent than generally thought.

Most "f" values in Table 3 lie between 0.10 and 0.25, even when nitrate was present at the surface in the equatorial upwelling of both the Pacific and Atlantic Oceans. Dugdale *et al.* (1992) found an overall mean "f" factor of 0.17 at 150°W (15°N to 15°S), which is in strong agreement with the median value calculated from all data reported in Table 3. This mean "f" = 0.17 seems to represent the typical magnitude of the tropical open ocean. Dugdale et al. (1992) observed the lowest "f" values in the nitrate-depleted waters (see their Fig. 3), but they did not find any clear relationship between "f" and the nitrate concentration in nutrient-rich waters. Similarly, neither "f" nor nitrate uptake were directly correlated to the nitrate concentration in the equatorial Atlantic, as long as ambient NO<sub>3</sub> exceeded about  $0.1 \mu M$ (Le Bouteiller, 1986). By contrast, total primary production and new production were strongly correlated, and "f" varied as a function of the chlorophyll concentration (Fig. 1). These relationships suggest that in the tropical open ocean, factors controlling the chlorophyll concentration are more important to consider for interpreting results of C and N production measurements than the NO<sub>3</sub> concentration. Chlorophyll, size, and composition of phytoplankton are related. In the equatorial Atlantic, Le Bouteiller and Herbland (1984) have shown that in waters containing  $1 \text{ mg m}^{-3}$  of total chlorophyll a, about 50% of chlorophyll belonged to organisms passing through a  $3 \mu m$  filter, and the percentage approximated 75% in oligotrophic waters. In addition, in stratified tropical situations, the euphotic zone

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Figure 1. (a) Primary production (<sup>14</sup>C) vs <sup>15</sup>N-NO<sub>3</sub> uptake. Results from four studies at the fixed position 0° 4°W in the Atlantic Occan; 24 days of *in situ* incubations (11 h) from dawn to dusk. CIPREA 4 cruise: all data (n = 41) are represented by only one point (mean  $\pm$  SD). Data from nitrate-poor waters (NO<sub>3</sub> <1  $\mu$ M) were excluded. Redrawn from Le Bouteiller (1986). (b) f-ratio (%) vs chlorophyll *a*. Same data set as in Figure 1 (a), including data from nitrate-depleted waters.

was found to be divided into two parts: an upper nitratedepleted layer in which cyanobacteria were always numerically predominant, closely linked with chorophyll *a* in the  $<1 \mu$ m fraction which accounted for 60% of total chorophyll *a* on average; a lower nutrient-rich layer in which chlorophyll  $>1 \mu$ m dominated, belonging mainly to eukaryotic microalgae (Le Bouteiller *et al.*, 1992). Size fractionation experiments have suggested that picophytoplankton have a preference for ammonium and the larger cells a preference for nitrate (Le Bouteiller, 1986). As a consequence, the new production ratio "f" would be directly related to the relative abundance of large cells able to take up nitrate (Fig. 1b), and the chlorophyll concentration would reflect the size structure of phytoplankton.

Field data (Table 2) show than on average,  $\rho C/\rho N$ actually equals C/N assessed in phytoplankton (Table 1). If we suppose that the assimilation ratio is indeed 6.6in most regions, then many data in Table 2 show that C production is up to twice N production or, inversely, four to five times lower than N production. Over- or underestimation of C or N production would be the cause of such great discrepancies. Further measurements of total C and N uptake rates are needed in the future in order to reduce such uncertainty and to enable calculation of N uptake from C production measurements. The number of nitrogenous sources taken up by phytoplankton, the necessity to obtain reliable data of NH<sub>4</sub> and urea concentrations in sea water (McCarthy, 1980, 1981), the correction of isotope dilution in <sup>15</sup>N-NH<sub>4</sub> and <sup>15</sup>N-urea utilization, and the conversion of short-term incubation measurements to daytime N assimilation, are some of the main difficulties to resolve in total N production estimation. Although never really simple, total production seems nevertheless much easier to obtain in terms of carbon than nitrogen. When the C/N assimilation ratio of phytoplankton is known in a system, then the simplest way to obtain "f" lies in measuring new production by the <sup>15</sup>N method and total production by the <sup>14</sup>C or the <sup>13</sup>C method.

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