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## A seasonal tropical sink for atmospheric CO<sub>2</sub> in the Atlantic ocean: the role of the Amazon River discharge

J.F. Ternon <sup>a</sup>, C. Oudot <sup>b,\*</sup>, A. Dessier <sup>b</sup>, D. Diverres <sup>b</sup>

<sup>a</sup> IRD, Cayenne, French Guiana

<sup>b</sup> IRD, Brest, France

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## A seasonal tropical sink for atmospheric CO<sub>2</sub> in the Atlantic ocean: the role of the Amazon River discharge

J.F. Ternon<sup>a</sup>, C. Oudot<sup>b,\*</sup>, A. Dessier<sup>b</sup>, D. Diverres<sup>b</sup>

<sup>a</sup> IRD, Cayenne, French Guiana

<sup>b</sup> IRD, Brest, France

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### Abstract

In the western equatorial Atlantic ocean, near-surface observations show that during summertime, the low-salinity oceanic water, arising from mixing with the Amazon River discharge at the equator, has low CO<sub>2</sub> fugacity levels. Near the coast of South America where the salinities are the lowest ( $S < 20$ ), the fugacity of oceanic CO<sub>2</sub> decreases down to 150  $\mu\text{atm}$  and the shelf area acts as a significant sink for atmospheric CO<sub>2</sub>. The dilution effect by low-salinity water only partly accounts for the decrease in CO<sub>2</sub>, and the biological production in the Amazon Plume water enriched in nutrients lowers dissolved inorganic carbon and decreases the  $f\text{CO}_2$  by nearly 30%. The low-salinity Amazon water tongue spreads northwestwards along the coast by the North Brazil Current (NBC) and is deflected eastwards north of 5°N in the NBC retroflexion in summer. Consequently, the low-salinity and oceanic  $f\text{CO}_2$  (below the atmospheric  $f\text{CO}_2$  level) signatures may extend more than 2000 km eastwards. The impact of the river outflow on the air–sea CO<sub>2</sub> exchanges in the western region is demonstrated by using the climatologies of the sea surface salinity (SSS) to estimate the magnitude of the annual net CO<sub>2</sub> flux in the western part of the equatorial Atlantic. This is in contrast with the central and eastern parts that are sources for atmospheric CO<sub>2</sub>. © 2000 Elsevier Science B.V. All rights reserved.

**Keywords:** surface seawater;  $f\text{CO}_2$ ; salinity; CO<sub>2</sub> flux; sink; river; productivity

### 1. Introduction

The Atlantic equatorial belt (10°N–10°S) is generally regarded as an oceanic source of CO<sub>2</sub> for the atmosphere (Keeling and Waterman, 1968; Takahashi et al., 1978; Roos and Gravenhorst, 1984; Smethie et al., 1985; Andrié et al., 1986; Oudot et

al., 1987; Oudot et al., 1995; Schneider and Morlang, 1995; Goyet et al., 1998a; Lefèvre et al., 1998), because of zonal spreading of the oversaturated equatorial water (equatorial upwelling of CO<sub>2</sub>-rich deep waters). Nevertheless, sink regions for atmospheric CO<sub>2</sub> may appear in the equatorial latitude band because precipitation and river run-off can affect the fugacity (or partial pressure) of surface water CO<sub>2</sub>,  $f\text{CO}_2$ . The largest rivers in the world discharge into the Atlantic ocean (Perry et al., 1996): the Amazon River in the west (0°, 50°W) and the

\* Corresponding author. Centre IRD (Ex ORSTOM), B.P. 70, 29280-Plouzané, France. Tel.: +33-2-98-22-45-01; Fax: +33-2-98-22-45-14; E-mail: oudot@ird.fr

Congo River in the east (6°S, 12°E). Previous papers showed how surface  $f\text{CO}_2$  can be significantly affected by river run-off (Kempe and Pegler, 1991; Chen, 1993; Bakker et al., 1996; Frankignoulle et al., 1996; Kumar et al., 1996). Freshwater input by rivers decreases the ocean surface salinity and reduces concentration of total inorganic  $\text{CO}_2$  and  $\text{CO}_2$  fugacity. Considering the large discharge of the Amazon River ( $1.93(\pm 0.01) \times 10^5 \text{ m}^3 \text{ s}^{-1}$ , as reported by Perry et al., 1996) and the retroflexion of the North Brazil Current (NBC) (Richardson and Reverdin, 1987), the resulting plume of low-salinity water influences a large portion of the western tropical Atlantic (Neumann, 1969; Muller-Karger et al., 1988; Dessier and Donguy, 1994). Using Coastal Zone Coastal Scanner (CZCS) climatologies, Lefèvre et al. (1998) discuss the eastward advection of the Amazon outflow by the North Equatorial Counter Current (NECC) as far as 25°W. Here, we present a data set that illustrates the effect of the Amazon run-off on the surface water  $f\text{CO}_2$  in the western tropical Atlantic. Then, we discuss the physical and biological processes controlling the spatial variations of  $f\text{CO}_2$  in that area, and finally, we estimate seasonal changes of the air–sea  $\text{CO}_2$  flux deduced from the seasonal cycle of the sea surface salinity (SSS) in the region.

## 2. Data set

The data used in this study were obtained during the two ETAMBOT cruises carried out in boreal summer (9 September 1995–11 October 1995) and boreal spring (12 April 1996–16 May 1996), respectively, in the western tropical Atlantic (Fig. 1a) and the SABORD survey (25–29 May 1996) of the continental shelf off French Guiana (Fig. 1b). The ETAMBOT cruises were part of the French contribution to the WOCE Hydrographic Programme in the Atlantic ocean. Data were collected during the WOCE CITHER 1 cruise (January–March 1993) in the same region, but the results are not used in this study because of possible bias in the CITHER 1  $f\text{CO}_2$  data set.

The hydrographic and  $\text{CO}_2$  data were collected in a similar fashion as during the CITHER 1 cruise carried out in the equatorial Atlantic ocean (Oudot et

al., 1995). Sea surface water was sampled with a General Oceanics rosette system fitted to a Neil Brown CTDO probe (Gouriou, 1997a,b).

The measurement of  $f\text{CO}_2$  in seawater was carried out by equilibrating an air stream with a thermostated (28.0°C) seawater sample (550 cm<sup>3</sup>) in the sampling flask itself. The equilibrated air was injected into the IR analyzer (LICOR, model LI 6262) after passing through a desiccant ( $\text{P}_4\text{O}_{10}$ ) column. The measured signal was compared with those produced by three standard gases (329.0–349.6–407.7 ppm), certified to  $\pm 0.25$  ppm by the French supplier (Air Liquide) and in agreement with the Scripps Institution of Oceanography (SIO) standard scale. The  $\text{CO}_2$  fugacity was then calculated following Weiss (1974) taking the vapor pressure of water into account (Weiss and Price, 1980) and corrected for in situ temperature according to the temperature dependence equation described by Copin-Montegut (1989). The reproducibility of measurements, tested as the mean difference for duplicates, is 2.9  $\mu\text{atm}$  during ETAMBOT 1 and 2.2  $\mu\text{atm}$  during ETAMBOT 2 and SABORD. Due to calibration anomalies at the beginning of the ETAMBOT 1 cruise, calculated  $f\text{CO}_2$  values (from  $\text{TCO}_2$  and pH) are reported rather than measured  $f\text{CO}_2$  data for the eight first stations of that cruise.

Atmospheric  $\text{CO}_2$  concentration (ppm) was measured twice a day by continuously pumping for about 30 min an air stream, taken at a mast at the vessel's bow (about 10 m above the sea surface). The dried air stream was introduced into the IR analyzer and calibrated with the same standard gases as described previously. The  $\text{CO}_2$  fugacity in the atmosphere ( $\mu\text{atm}$ ) was then computed taking the water vapor pressure and the barometric pressure into account. The uncertainty for the atmospheric  $\text{CO}_2$  concentration is estimated at  $\pm 0.3$  ppm ( $\pm 0.3 \mu\text{atm}$  for the atmospheric  $\text{CO}_2$  fugacity).

Measurements of total inorganic carbon,  $\text{TCO}_2$ , were made by gas chromatography (Oudot et al., 1995) and their reproducibility, determined as the mean difference of duplicates (two bottles fired at the same depth), is 7  $\mu\text{mol kg}^{-1}$  during ETAMBOT 1 and 10  $\mu\text{mol kg}^{-1}$  during ETAMBOT 2 and SABORD. The accuracy was checked against Certified Reference Material provided by A.G. Dickson (Scripps Institution of Oceanography).

The total alkalinity of seawater, TA, was determined from  $\text{TCO}_2$  and pH measurements using the UNESCO (1987) definition:

$$\text{TA} = [\text{HCO}_3^-] + 2[\text{CO}_3^{2-}] + [\text{B(OH)}_4^-] + [\text{OH}^-] - [\text{H}^+]. \quad (1)$$

$[\text{HCO}_3^-]$  and  $[\text{CO}_3^{2-}]$  were deduced from  $\text{TCO}_2$  measurements and pH measurements. The pH measurements were based on the total hydrogen ion concentration scale (Dickson, 1993), and the Ross combination electrode (ORION, model 81-02) was calibrated in 0.04 M Tris and 0.04 M AMP buffers prepared according to Dickson (1993). The reproducibility of pH measurements, tested as the mean difference of duplicates is 0.002 and 0.003 during ETAMBOT 1 and ETAMBOT 2, respectively. The equations used for determination of the dissociation constants relative to the 'total' pH scale were those of Roy et al. (1993) for carbonic acid, Dickson (1990) for boric acid and Millero (1995) for ionization of water. The reproducibility of TA results, tested as the mean difference of duplicates, is similar to that of  $\text{TCO}_2$  measurements:  $7 \mu\text{mol kg}^{-1}$  during ETAMBOT 1 and  $10 \mu\text{mol kg}^{-1}$  during ETAMBOT 2 and SABORD.

The nutrient (silicate) concentrations were measured according to the standard method (Mullin and Riley, 1955) in a BRAN + LUEBBE Auto-Analyzer equipped with an automated data acquisition system. The analytical precision ( $\pm 1$  SD) was  $0.2 \mu\text{mol kg}^{-1}$  for silicate (Baurand and Oudot, 1997a,b) for each cruise.

### 3. Results: $f\text{CO}_2$ distributions

$f\text{CO}_2$  data obtained along the cruise track during ETAMBOT 1 and ETAMBOT 2 are shown in Fig. 2, together temperature and salinity. The mean value of atmospheric  $\text{CO}_2$  fugacity during both cruises is depicted as straight lines for ETAMBOT 1 ( $347.0 \mu\text{atm}$  — full line) and ETAMBOT 2 ( $352.5 \mu\text{atm}$  — dashed line) to illustrate the  $\text{CO}_2$  sink (oceanic  $f\text{CO}_2 < \text{atmospheric } f\text{CO}_2$ ) and source (oceanic  $f\text{CO}_2 > \text{atmospheric } f\text{CO}_2$ ) areas. Near the coast, at the beginning of ETAMBOT 1 and the end of ETAMBOT 2, the oceanic  $f\text{CO}_2$  decreases down to

$200 \mu\text{atm}$  (i.e.,  $150 \mu\text{atm}$  below the atmospheric level), while, in the open ocean the highest  $f\text{CO}_2$  level remains below  $400 \mu\text{atm}$  (i.e., only  $50 \mu\text{atm}$  above the atmospheric level). Results collected during the SABORD survey are shown in the same way in Fig. 3: the continental shelf is a strong  $\text{CO}_2$  sink everywhere with the ocean–atmosphere  $f\text{CO}_2$  difference reaching more than  $200 \mu\text{atm}$ .

In both figures, spatial variations of oceanic  $f\text{CO}_2$  are closely related to those of salinity. During ETAMBOT 1, which was carried out during late summer–early fall, the salinity field is strongly influenced by the Amazon Plume water transported offshore in the NBC retroflection (Lentz, 1995). The low-salinity ( $< 33.0$ ) and low-fugacity ( $\sim 320 \mu\text{atm}$ ) water extends eastwards to about  $35^\circ\text{W}$ , over  $2000 \text{ km}$  from the river mouth (Fig. 2). The eastward extension of the low-salinity plume is not continuous due to the influence of the NBC retroflection eddies and the meandering of the NECC (Richardson et al., 1994; Bourlès et al., 1999). During ETAMBOT 2 that started in April, a few weeks before the maximum Amazon discharge (Lentz, 1995), the SSS ( $\sim 36.0$ ) as well as  $f\text{CO}_2$  (on the average  $\sim 360 \mu\text{atm}$ ) are generally high, except in front of the Amazon mouth (start of leg D, Fig. 2b) and in front of Cayenne at the end of the cruise, in mid-May (leg E, Fig. 2b).

On the French Guiana continental shelf (SABORD), both the salinity and  $f\text{CO}_2$  values increase from inshore to offshore along each leg at the end of May (Fig. 3). There is also an increasing trend of salinity and  $f\text{CO}_2$  northwestwards along the coast, i.e., downstream the plume carried by the northwestward boundary current. The very fresh waters encountered during the cruise, as well as those identified two weeks earlier (ETAMBOT 2 — leg E) in front of Cayenne, probably correspond to relatively pure Amazon water that also exhibits the most pronounced  $f\text{CO}_2$  signature. High silicate concentrations (up to  $45 \mu\text{mol kg}^{-1}$ ) measured on the continental shelf confirm the riverine origin of these waters.

No clear temperature– $f\text{CO}_2$  relationship is evidenced in the oceanic  $\text{CO}_2$  fugacity distributions. This absence of relationship was previously shown (Oudot et al., 1987) in tropical Atlantic, except north of  $5^\circ\text{N}$  in boreal winter where surface-layer cooling

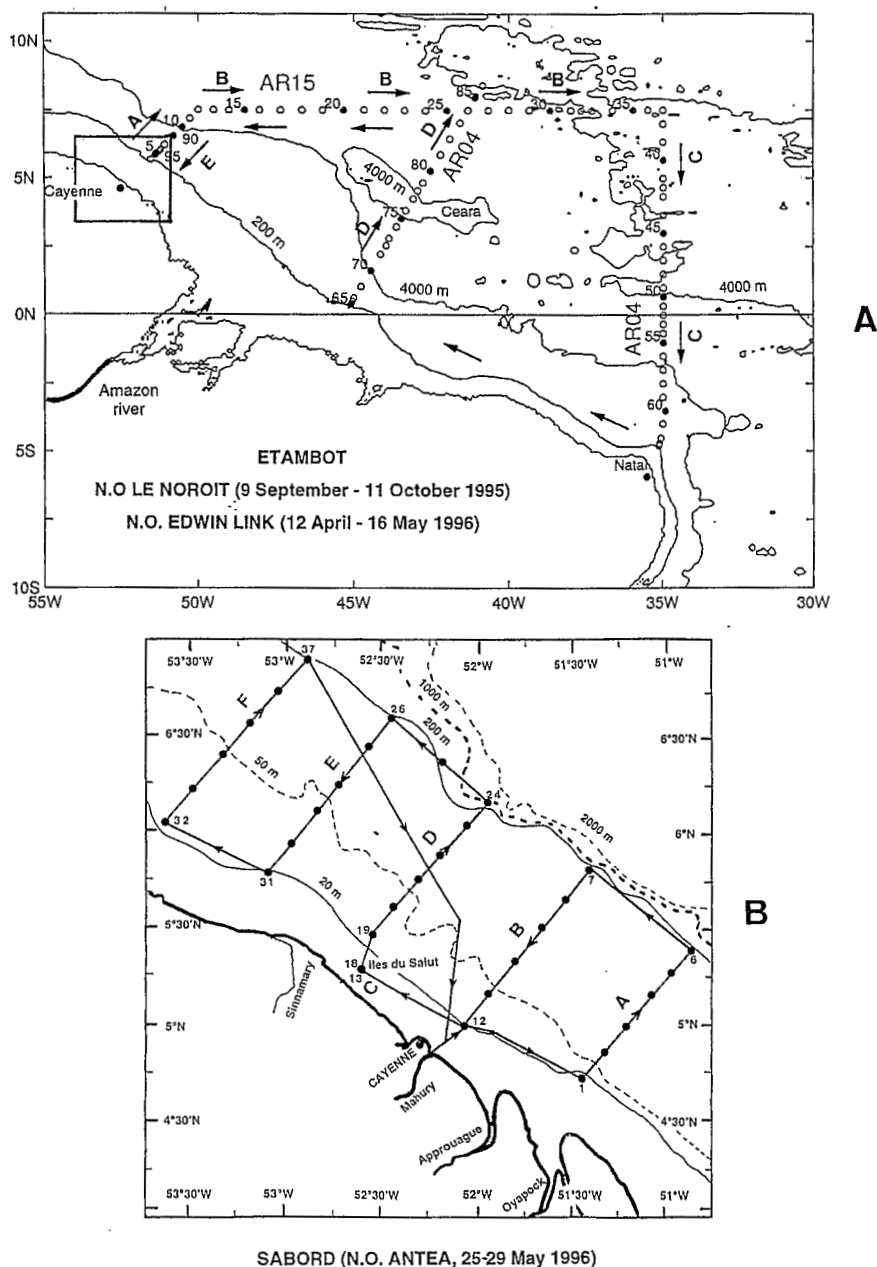


Fig. 1. Cruise tracks and stations location during (A) ETAMBOT 1 (9 September 1995–11 October 1995) and ETAMBOT 2 (12 April 1996–16 May 1996), and (B) SABORD (25–29 May 1996). The square in (A) indicates the position of (B).

may lower the oceanic  $f\text{CO}_2$  value. However, during ETAMBOT 2 which is made just in early spring, the slight decrease of oceanic  $f\text{CO}_2$  below the atmospheric level, at the end of leg B (Fig. 2c), may be related to the observed temperature decrease (Fig.

2a). Thus, oceanic  $f\text{CO}_2$  corrected at a constant temperature (28°C), according to the  $\text{CO}_2$ -solubility temperature dependence of  $4.23\% \text{ } ^\circ\text{C}^{-1}$  for constant total  $\text{CO}_2$  (Takahashi *et al.*, 1993), varies much less and remains above the atmospheric level during

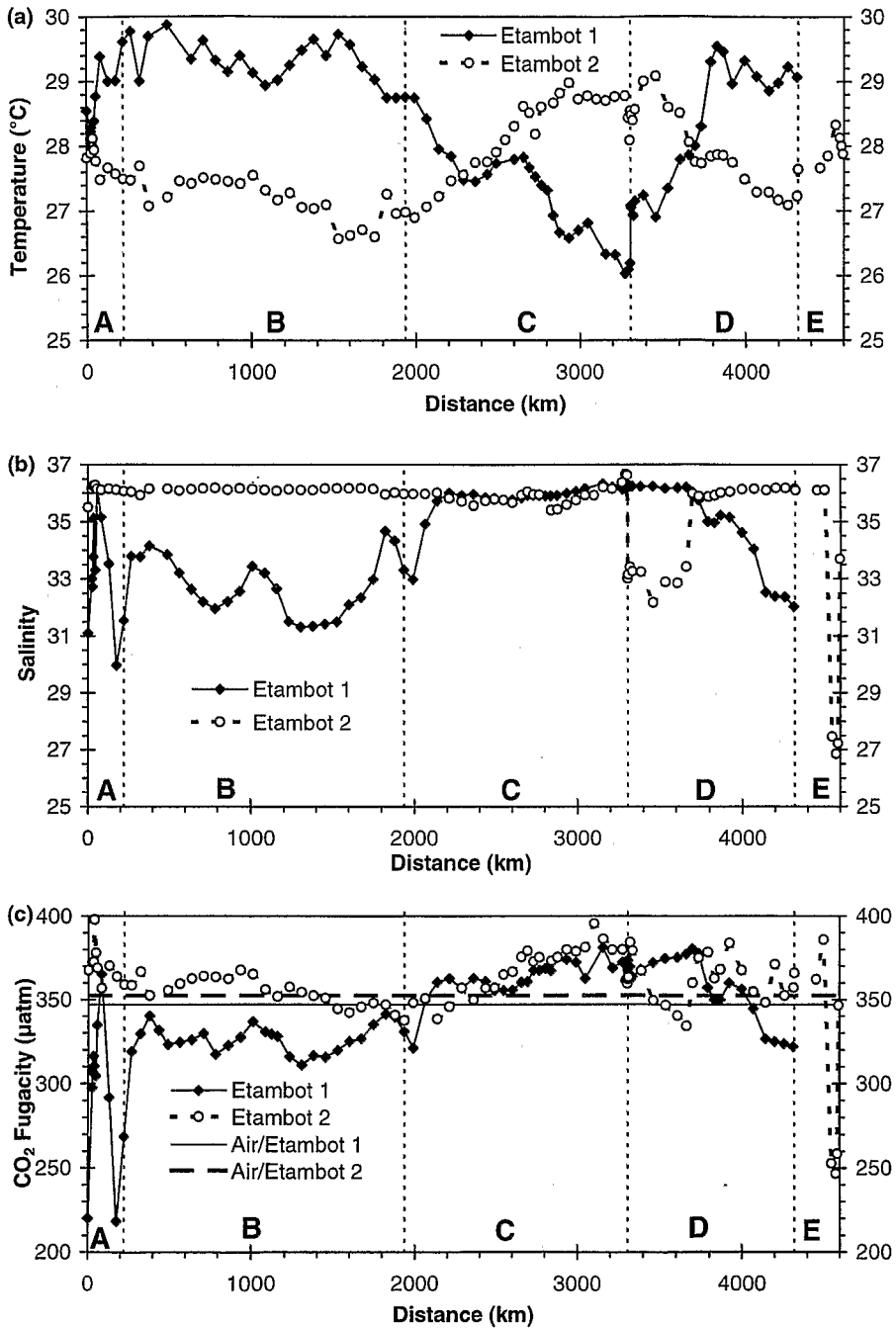


Fig. 2. Variations of (a) temperature (°C), (b) salinity and (c) fugacity of CO<sub>2</sub> (µatm) at the sea surface along the cruise tracks of ETAMBOT 1 and ETAMBOT 2. Mean values of atmospheric CO<sub>2</sub> fugacity are indicated by straight lines (347.0 µatm for ETAMBOT 1 and 352.5 µatm for ETAMBOT 2). Letters correspond to the different legs of the cruises (Fig. 1a). The CO<sub>2</sub> fugacity data of ETAMBOT 1 on legs A and B for the first 600 km of the ship's route are calculated from TCO<sub>2</sub> and pH, and are omitted in Fig. 5a.

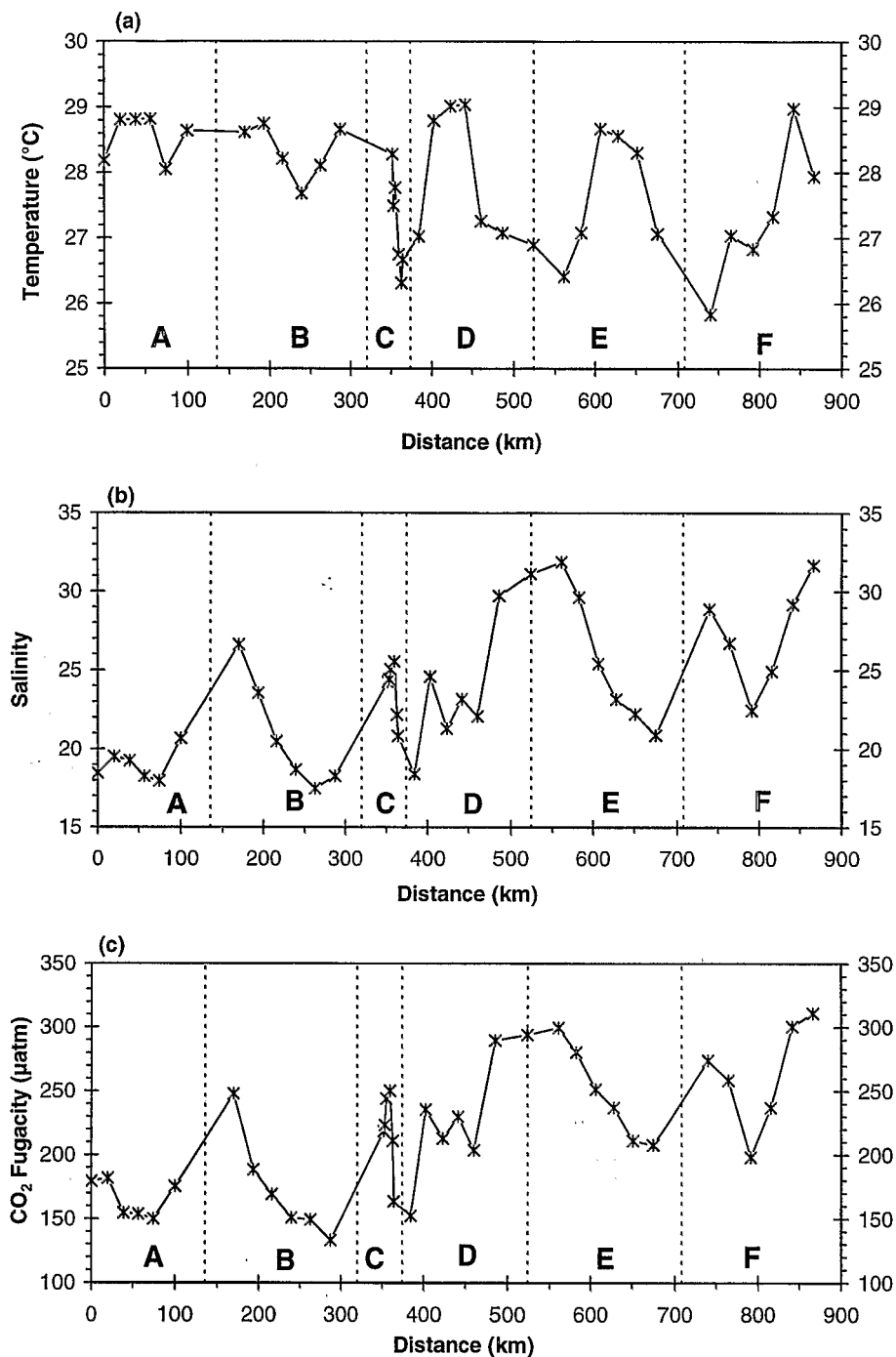


Fig. 3. Variations of (a) temperature (°C), (b) salinity and (c) fugacity of CO<sub>2</sub> (µatm) at the sea surface along the SABORD cruise track. Letters correspond to the different legs of the survey (Fig. 1). Mean value of atmospheric CO<sub>2</sub> fugacity (not shown) is 352.5 µatm as determined during ETAMBOT 2.



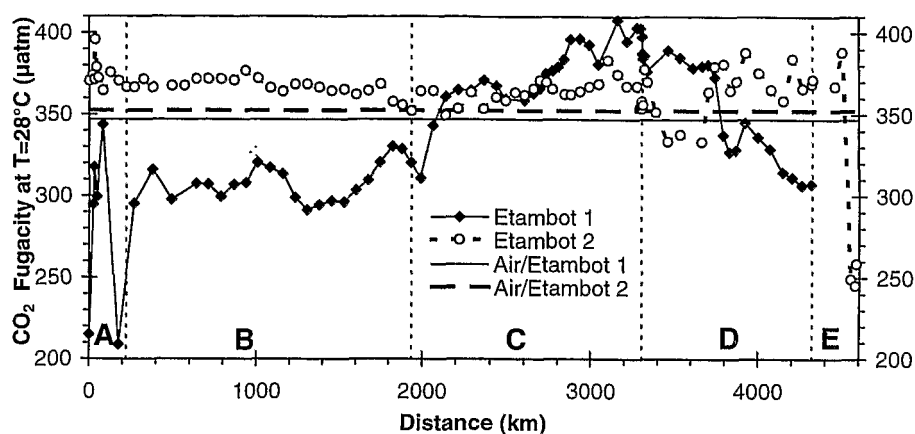


Fig. 4. Variations of fugacity of  $\text{CO}_2$  ( $\mu\text{atm}$ ) measured at  $T=28^\circ\text{C}$  at the sea surface along the cruise tracks of ETAMBOT 1 and ETAMBOT 2. Mean values of atmospheric  $\text{CO}_2$  fugacity are indicated by straight lines ( $347.0 \mu\text{atm}$  for ETAMBOT 1 and  $352.5 \mu\text{atm}$  for ETAMBOT 2). Letters correspond to the different legs of the cruises (Fig. 1a).

ETAMBOT 2 (Fig. 4), except where salinity strongly decreases.

#### 4. Discussion: physical and biological processes for $\text{CO}_2$ sink

As expected from the similar distribution patterns of  $f\text{CO}_2$  and salinity in surface waters (Figs. 2 and 3), we found a high correlation ( $r = 0.979$ ) between  $\text{CO}_2$  fugacity and salinity at the sea surface in the western equatorial region (Fig. 5a), irrespective of the season. That relationship between  $f\text{CO}_2$  and salinity suggests that the decrease in oceanic  $f\text{CO}_2$  below the atmospheric level is closely related to Amazon discharge and the subsequent spreading of freshwater by the surface oceanic circulation. Such a salinity decrease in the area cannot be the result of precipitation: the rainfall map of Dorman and Bourke (1981), as well as the freshwater budget of Yoo and Carton (1990) in the tropical Atlantic ocean, suggest that the influence of precipitation is negligible in comparison to the Amazon River discharge. On the other hand, the strong inverse correlation between salinity and silicate (Fig. 5b) strongly suggests that the decrease in salinity is related to the river discharge (Froelich et al., 1978; Key et al., 1985).

However, processes other than physical ones (mixing between ocean water and river freshwater,

precipitation, temperature decrease) contribute to lower  $f\text{CO}_2$ . Biological activity also may decrease the oceanic  $\text{CO}_2$  fugacity by photosynthetic  $\text{CO}_2$  uptake. It is possible to estimate the dilution effect of freshwater on the  $f\text{CO}_2$  distribution by computing  $f\text{CO}_2$  expected from the mixing between ocean water and river water. Expected  $f\text{CO}_2$  is calculated (dotted line in Fig. 5a), at a given salinity and the mean sea surface temperature (SST) in the region ( $28^\circ\text{C}$ ), from  $\text{TCO}_2$  and TA values resulting from the conservative mixing between the two endmembers. According to the ETAMBOT data set (Fig. 5c–d), the seawater component values are:

$$\text{Ocean water: } S = 36.2 \quad \text{TCO}_2 = 2018 \mu\text{mol kg}^{-1}$$

$$\text{TA} = 2403 \mu\text{mol kg}^{-1} \quad (2)$$

To estimate the Amazon River endmember, we refer to Richey et al. (1990) which indicate a  $\text{TCO}_2$  concentration of  $500\text{--}600 \mu\text{mol kg}^{-1}$  downstream (in a more recent study, Richey et al., 1991 mention a range of  $485\text{--}667 \mu\text{mol kg}^{-1}$  at the most down-river site); the  $\text{TCO}_2$  values were slightly greater during rising water (March) than falling water (September). We neglect the decomposition of organic carbon of which the concentration is comparable to inorganic carbon, because a large proportion of the organic matter is refractory (Richey et al., 1990). As the hydrogen carbonate is the predominant form of

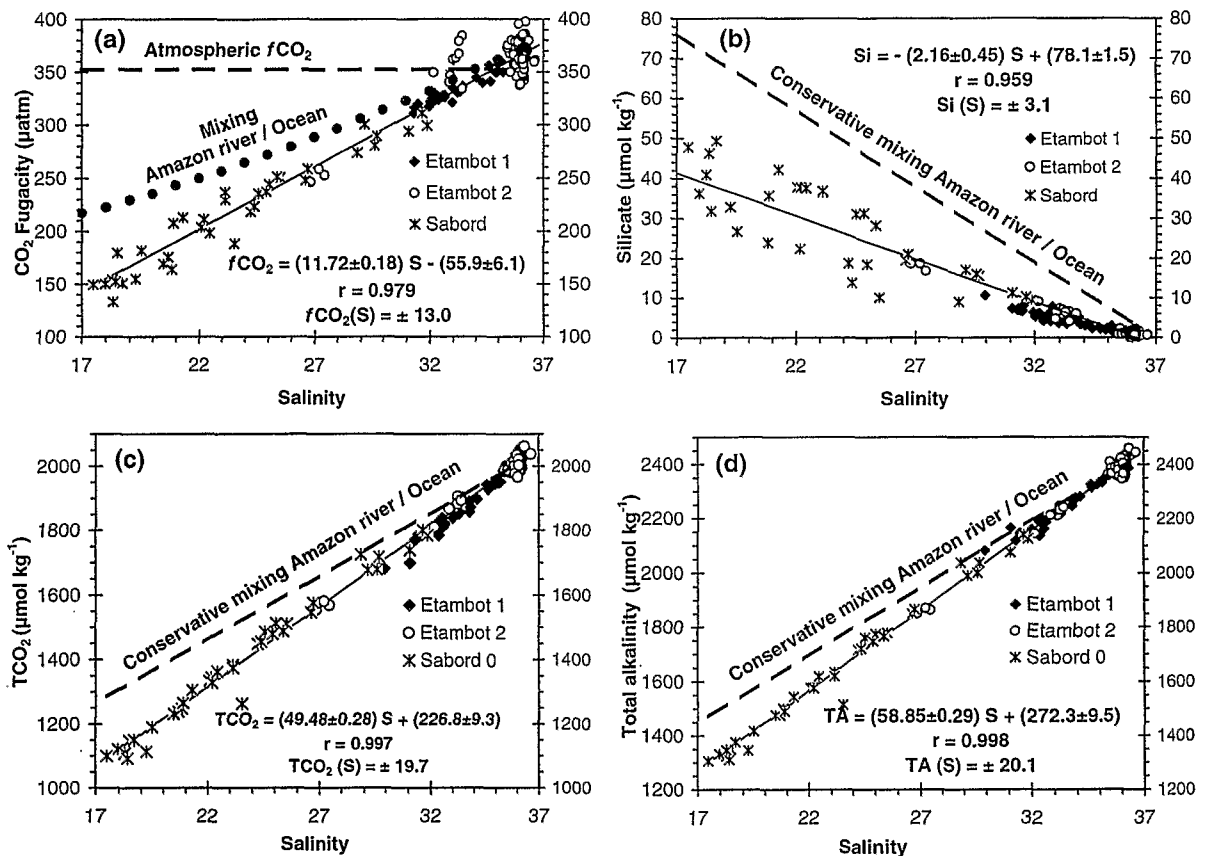


Fig. 5. (a) Relationship between  $\text{CO}_2$  fugacity ( $\mu\text{atm}$ ) and salinity at the sea surface during ETAMBOT 1, ETAMBOT 2 and SABORD. The equation and correlation coefficient,  $r$ , are for the regression line (solid line) fitted to the data.  $f\text{CO}_2(S)$  indicates the uncertainty in  $f\text{CO}_2$  calculated from salinity– $f\text{CO}_2$  regression equation. The dotted line represents expected  $f\text{CO}_2$  from the mixing between ocean water and river water: expected  $f\text{CO}_2$  is calculated at a given salinity and the mean sea surface temperature in the region ( $28^\circ\text{C}$ ) from  $\text{TCO}_2$  and TA values resulting from the conservative mixing between river and sea waters, whose endmembers are defined in the text. (b) Relationship between silicate ( $\mu\text{mol kg}^{-1}$ ) and salinity at the sea surface during ETAMBOT 1, ETAMBOT 2 and SABORD. The equation and correlation coefficient,  $r$ , are for the regression line (solid line) fitted to the data.  $\text{Si}(S)$  indicates the uncertainty in silicate calculated from salinity–silicate regression equation. The dashed line represents the conservative mixing between river and sea waters, whose endmembers are defined in the text. (c) Relationship between total inorganic carbon  $\text{TCO}_2$  ( $\mu\text{mol kg}^{-1}$ ) and salinity at the sea surface during ETAMBOT 1, ETAMBOT 2 and SABORD. The equation and correlation coefficient,  $r$ , are for the regression line (solid line) fitted to the data.  $\text{TCO}_2(S)$  indicates the uncertainty in  $\text{TCO}_2$  calculated from salinity– $\text{TCO}_2$  regression equation. The dashed line represents the conservative mixing between river and sea waters, whose endmembers are defined in the text. (d) Relationship between total alkalinity TA ( $\mu\text{mol kg}^{-1}$ ) and salinity at the sea surface during ETAMBOT 1, ETAMBOT 2 and SABORD. The equation and correlation coefficient,  $r$ , are for the regression line (solid line) fitted to the data.  $\text{TA}(S)$  indicates the uncertainty in TA calculated from salinity–TA regression equation. The dashed line represents the conservative mixing between river and sea waters, whose endmembers are defined in the text.

the inorganic carbon in the Amazon water (Richey et al., 1990), the river endmember value of TA may be considered to be equal to the  $\text{TCO}_2$  value, as mentioned by Olsson and Anderson (1997). The choice of the  $\text{TCO}_2$  and TA values in Amazon River water will govern the depression of  $f\text{CO}_2$  by mixing. As

the Sabord survey occurred in May, we adopt the  $\text{TCO}_2$  and TA values observed during rising water and the river water endmembers are:

$$\text{River water: } S=0 \quad \text{TCO}_2=600 \mu\text{mol kg}^{-1}$$

$$\text{TA} = 600 \mu\text{mol kg}^{-1}. \quad (3)$$

For a salinity of 17.5,  $\text{TCO}_2$  and TA expected from the conservative mixing are 1285 and 1472  $\mu\text{mol kg}^{-1}$ , respectively, and the corresponding calculated  $f\text{CO}_2$  value is 219  $\mu\text{atm}$ . As the mean observed  $\text{CO}_2$  fugacity of the ocean water ( $S = 36.2$ ) is about 368  $\mu\text{atm}$  (Fig. 5a), the maximal depression in  $f\text{CO}_2$  expected from the mixing effect is 149  $\mu\text{atm}$ , i.e., about 67% of the maximal observed  $f\text{CO}_2$  decrease (220  $\mu\text{atm}$ ). The biological processes should therefore account for nearly third of the observed depression in oceanic  $f\text{CO}_2$ . If we used lower  $\text{TCO}_2$  and TA values of 500  $\mu\text{mol kg}^{-1}$  in river water, the expected  $f\text{CO}_2$  will be lowered by 16  $\mu\text{atm}$  and the biological depression would be yet a quarter of the observed depression in oceanic  $f\text{CO}_2$ , that does not alter our conclusions.

Some of the ETAMBOT 2 data, with salinity around 33, are closer to the mixing line than to the regression line (Fig. 5a). They correspond to samples, collected in front of the Amazon estuary, for which the biological uptake of  $\text{CO}_2$  should be reduced, probably because of less favorable conditions of biological production (greater turbidity).

The biological activity effect (photosynthesis/respiration or carbonate production/dissolution) is also visible on the plots of total inorganic carbon  $\text{TCO}_2$  and total alkalinity TA vs. salinity (Fig. 5c–d), and silicate vs. salinity (Fig. 5b). As for  $f\text{CO}_2$  data, the  $\text{TCO}_2$ , TA and silicate fall below the conservative mixing lines. The mixing lines for  $\text{TCO}_2$  and TA (Fig. 5c–d) are drawn from the endmember values previously defined in (2) and (3). The mixing line for silicate (Fig. 5b) is constructed from Amazon River estuary surface samples (Table 2 in Key et al., 1985) and from mean oceanic concentration determined during our cruises, resulting in the following endmembers:

$$\text{Ocean water: } S = 36.2 \quad \text{Silicate} = 2 \mu\text{mol kg}^{-1} \quad (4)$$

$$\text{River water: } S = 0 \quad \text{Silicate} = 142 \mu\text{mol kg}^{-1}. \quad (5)$$

DeMaster and Pope (1996) found 144  $\mu\text{mol kg}^{-1}$  as silicate concentration in the riverine endmember, based on AMASSEDS results.

The silica data, as well as those of  $\text{TCO}_2$  and TA, illustrate the biological removal of carbon and silica

in sea surface water by calcifying and siliceous organisms. From the maximum  $\text{TCO}_2$  deficit recorded in the SABORD data, which is about 200  $\mu\text{mol kg}^{-1}$  (Fig. 5c), we can estimate the primary production rate. Lentz and Limeburner (1995) give an averaged low-salinity layer thickness of  $5 \pm 2$  m within the Amazon Plume, which rarely exceeds 10 m. The thickness of the low-salinity layer on the continental shelf, in front of French Guiana, ranges from 10 to 15 m during SABORD, while the depth of the surface mixed layer is 15 to 20 m, based on a study of the barrier layer in the western equatorial Atlantic (Pailler et al., 1999). From these considerations, we therefore assume that the  $\text{CO}_2$  biological consumption occurs in a  $10 \pm 5$  m thick layer. Lentz and Limeburner (1995) estimated the mass and salinity budget of the plume northwestward from the Amazon mouth, based on an along-plume velocity of 40  $\text{cm s}^{-1}$ . From this crude estimate, we predict that biological uptake of  $\text{CO}_2$  occurred during the water mass transit of 20 days from the river mouth to the French Guiana shelf, which is probably an upper limit as deduced from a tracer budget. With such a residence time, the rate of productivity rate is:

$$\begin{aligned} & [200 * (10 \pm 5)] / [20] \\ & = 100.0 \pm 72.5 \text{ mmol C m}^{-2} \text{ day}^{-1} \\ & = 1.2 \pm 0.9 \text{ g C m}^{-2} \text{ day}^{-1}. \end{aligned} \quad (6)$$

On the other hand, direct current measurements (satellite drifter observations according to Limeburner et al., 1995, moored array according to Lentz and Limeburner, 1995 and Acoustic Doppler Current Profiler according to Bourlès, personal communication) described the near-surface velocity fields over the North Brazil and Guiana shelves, respectively. These measurements lead us to estimate a lower transit time along the continental shelf of about 15 days, using a mean velocity of 50  $\text{cm s}^{-1}$  between the river mouth and 6°N. With this transit time, the productivity rate amounts to:

$$\begin{aligned} & [200 * (10 \pm 5)] / [15] \\ & = 133.3 \pm 96.7 \text{ mmol C m}^{-2} \text{ day}^{-1} \\ & = 1.6 \pm 1.2 \text{ g C m}^{-2} \text{ day}^{-1}. \end{aligned} \quad (7)$$

These estimates have to be corrected for the air–sea flux of  $\text{CO}_2$  as, the continental shelf area

being a strong CO<sub>2</sub> sink, the resulting entering flux of atmospheric CO<sub>2</sub> into the sea underestimates the depression of *f*CO<sub>2</sub> due to the biological productivity. The net air–sea flux is estimated to be  $-15 \text{ mmol m}^{-2} \text{ day}^{-1}$  (i.e.,  $-0.18 \text{ g C m}^{-2} \text{ day}^{-1}$ ) or  $-24 \text{ mmol m}^{-2} \text{ day}^{-1}$  (i.e.,  $-0.29 \text{ g C m}^{-2} \text{ day}^{-1}$ ) as described below (the two results corresponding to two different calculation models). This correction (0.2 to 0.3 g C m<sup>-2</sup> day<sup>-1</sup>) is negligible in comparison with the uncertainty of our estimation of the productivity rate expected from the CO<sub>2</sub> deficit.

The rough estimate of the primary production, which is responsible of third of the *f*CO<sub>2</sub> depression, is compared with the assessment of the primary productivity from <sup>14</sup>C-uptake experiments on the continental shelf within the plume of the Amazon River, during AMASSEDS (Smith and DeMaster, 1996). The mean productivity ranges from 2.2 in turbid nutrient rich water to 0.8 g C m<sup>-2</sup> day<sup>-1</sup> in clear offshore regions. Cadee (1975) estimated a mean primary production rate of 0.9 g C m<sup>-2</sup> day<sup>-1</sup> in the 20–60 m depth region of the French Guiana continental shelf, which was five times that measured offshore. Our results ( $1.6 \pm 1.2$  and  $1.2 \pm 0.9 \text{ g C m}^{-2} \text{ day}^{-1}$ ) are in reasonable agreement with the upper and lower limits of direct measurements of primary productivity in the Amazon estuary and in the zone directly influenced by the Amazon river plume, given the crudeness of the estimates. Thus, the biological activity in the surface water affected by the Amazon discharge contributes significantly to the CO<sub>2</sub> sink that seasonally appears in the western equatorial Atlantic ocean.

From comparison between the regression line fitted to data and the conservative mixing line, the maximum depression in silicate (Fig. 5b) due to biological activity is approximately 35 μmol kg<sup>-1</sup>, which is in agreement with the highest silicate depletion (over 30 μmol kg<sup>-1</sup>) reported by DeMaster and Pope (1996) on the Amazon shelf. To compare silicate and inorganic carbon depletion involved in biological uptake (maximum depression of 35 and 200 μmol kg<sup>-1</sup>, respectively), we have to correct the inorganic carbon depletion for calcification. The observed TA deviation away from the conservative line (172 μmol kg<sup>-1</sup> for *S* = 17.5, Fig. 5d) is characteristic of calcium carbonate production in seawater (Broecker and Peng, 1982). As the change in TA

due to carbonate mineral production is twice as large as the change in TCO<sub>2</sub> (Skirrow, 1975) and by neglecting the small changes of TA due to production and decay of organic matter (Brewer and Goldman, 1976), the organic carbon production corrected for calcification is  $200 - 172/2 = 114 \text{ μmol kg}^{-1}$ . Thus, silicate and inorganic uptake are in a molar ratio Si/C = 0.31, in the range (0.15–0.4) of the molar silicon/carbon production ratio reported by DeMaster et al. (1996) in the waters of the Amazon shelf. This relatively high value (0.31) in comparison to the mean of 0.13 for low-latitude diatoms (Brzezinski, 1985) underlines the dominant role of diatoms in the primary production in this zone.

## 5. Air–sea fluxes of CO<sub>2</sub>

### 5.1. Flux calculation procedure

Air–sea fluxes are estimated from the observed *f*CO<sub>2</sub> data and the relationship:

$$F = k_t \alpha \Delta f\text{CO}_2 \quad (8)$$

where *k<sub>t</sub>* is the CO<sub>2</sub> gas transfer velocity, *α* is CO<sub>2</sub> solubility in seawater (Weiss, 1974) and Δ*f*CO<sub>2</sub> is the difference between the CO<sub>2</sub> fugacity in the ocean and that in the atmosphere (Δ*f*CO<sub>2</sub> = *f*CO<sub>2</sub><sup>sea</sup> – *f*CO<sub>2</sub><sup>air</sup>). The CO<sub>2</sub> flux is positive when CO<sub>2</sub> escapes from the ocean into the atmosphere and negative otherwise. While *α* and Δ*f*CO<sub>2</sub> are easily calculated, *k<sub>t</sub>* is difficult to estimate because of its dependence on several factors (wind speed, sea surface roughness, bubbles, thermal skin effect). Different ways have been shown to compute this coefficient from wind speed, and take these effects into account (Broecker and Siems, 1984; Liss and Merlivat, 1986; Tans et al., 1990; Watson et al., 1991; Robertson and Watson, 1992; Wanninkhof, 1992; Van Scoy et al., 1995; McNeil and Merlivat, 1996). We chose to compute the air–sea CO<sub>2</sub> fluxes with two relationships: that of Liss and Merlivat (1986) (thereafter L&M86), we used in our previous studies (Andrié et al., 1986; Oudot et al., 1995) but often suspected to yield low fluxes, and the more recent relationship of Wanninkhof (1992) (thereafter Wkf92), that yields higher fluxes (Bakker et al., 1997) and whose gas flux parameterization for climatological winds is in

best agreement with observations (Haines et al., 1997). Considering the uncertainty in the relationships between wind speed and air–sea gas exchange, we ignored the thermal skin effect on the air–sea CO<sub>2</sub> flux (Van Scoy et al., 1995).

Liss and Merlivat (1986) equations for different wind regimes are:

$$k_{20} = 0.17V \quad \text{for } 0 < V < 3.6 \text{ m s}^{-1} \quad (9)$$

$$k_{20} = 2.85V - 9.65 \quad \text{for } 3.6 < V < 13.0 \text{ m s}^{-1} \quad (10)$$

where  $k_{20}$  is the transfer velocity (or exchange coefficient) for CO<sub>2</sub> at 20°C expressed in cm h<sup>-1</sup> and  $V$  is the wind speed at 10 m above the sea level, expressed in m s<sup>-1</sup>. The wind data we used to compute the transfer velocity are extracted from an atlas of the tropical Atlantic wind stress climatology (2° × 2° gridded monthly mean, Servain et al., 1987) constructed from measurements made aboard volunteer observing ships (Bourlès and Marin, 1997; Marin, 1997). For the whole continental shelf, we adopted the average wind speed measured at sea during the cruise, which is very similar to the climatological wind. The exchange coefficient depends on the temperature:

$$k_t = k_{20}(Sc_t/600)^{-2/3} \quad \text{for } 0 < V < 3.6 \text{ m s}^{-1} \quad (11)$$

$$k_t = k_{20}(Sc_t/600)^{-1/2} \quad \text{for } 3.6 < V < 13.0 \text{ m s}^{-1} \quad (12)$$

where  $Sc_t$ , the Schmidt number for CO<sub>2</sub> at  $t$ °C, is determined using the polynomial relationship of Wkf92:

$$Sc_t = 2073.1 - 125.62t + 3.6276t^2 - 0.043219t^3. \quad (13)$$

Wkf92 proposed two relationships to calculate the gas transfer velocity: one for short-term wind speed data (shipboard measurements, scatterometer and radiometer data) and another one for long-term averaged wind speed data (related to <sup>14</sup>C invasion):

$$k_{st} = 0.31V_{st}^2(660/Sc_t)^{1/2} \quad \text{for short-term wind} \quad (14)$$

$$k_{lt} = 0.39V_{lt}^2(660/Sc_t)^{1/2} \quad \text{for long-term wind} \quad (15)$$

where  $k_{st}$  and  $k_{lt}$  are expressed in cm h<sup>-1</sup> and  $V_{st}$  and  $V_{lt}$  are expressed in m s<sup>-1</sup>. The Schmidt number of CO<sub>2</sub> in seawater at temperature = 20°C is 660 (Wanninkhof, 1992).

## 5.2. Flux results

Fig. 6 shows the distribution of CO<sub>2</sub> fluxes calculated according to the Wkf92 parameterization for short-term wind (Eq. (14)), from  $f\text{CO}_2$  data and wind speed along both ETAMBOT cruises tracks and along the SABORD one. The net CO<sub>2</sub> fluxes along each leg of the ETAMBOT cruises and on the French Guiana continental shelf, calculated according to the two parameterizations (L&M86 and Wkf92), are compared in Table 1. These results show the lower CO<sub>2</sub> fluxes calculated according to the L&M86 parameterization in comparison with the Wkf92 CO<sub>2</sub> fluxes (Bakker et al., 1997; Goyet et al., 1998b). On average, the L&M86 fluxes differ from the Wkf92 ones by 80%, but differences may be beyond 130% (Table 1). At 7.5°N (Leg A + Leg B), the mean CO<sub>2</sub> flux is entering into the sea during ETAMBOT 1 and escaping during ETAMBOT 2, except at the end of that latter cruise (Leg E) when it becomes a strong invasion, because of the  $f\text{CO}_2$  decrease associated to the freshwater pool. Along the 35°W and Ceara sections (Leg C and Leg D, Fig. 1, top panel), CO<sub>2</sub> escapes from the ocean during both ETAMBOT cruises (Table 1). The amplitude of CO<sub>2</sub> flux variations is higher during ETAMBOT 2 than during ETAMBOT 1 (Fig. 6) as the wind speed north of the equator is higher in boreal spring (ETAMBOT 2) than in late summer–fall (ETAMBOT 1), as a result of the north–south migration of the Intertropical Convergence Zone during the year. The CO<sub>2</sub> sink observed in boreal spring 2000 km away from the coast, in open ocean (35–40°W) at 7.5°N (Fig. 6, ETAMBOT 2) is caused by winter cooling of the sea surface (Oudot et al., 1987). The CO<sub>2</sub> sink observed during ETAMBOT 1 along the 7.5°N section (Leg B) is smaller than the previously described one (ETAMBOT 2) due to lower wind speed in late summer–fall and is related to the eastward spreading of low-salinity water (Fig. 2b). The strong CO<sub>2</sub> sink observed at the end of ETAMBOT 2, which was not present 1 month earlier (Fig. 6, ETAMBOT 2) or a few days later on the continen-

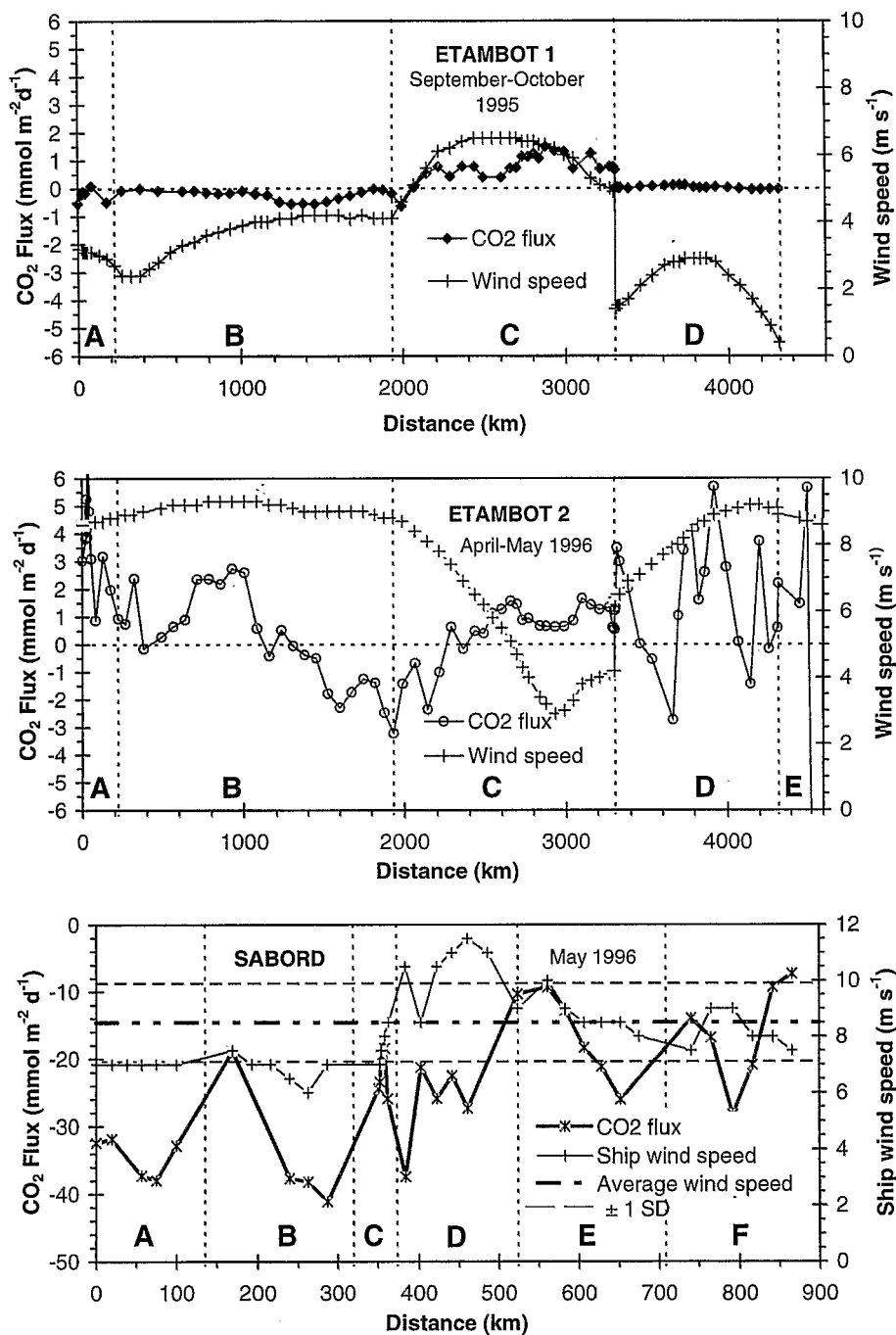


Fig. 6. Variations of CO<sub>2</sub> flux (mmol m<sup>-2</sup> day<sup>-1</sup>) at the air–sea interface and wind speed (m s<sup>-1</sup>) along cruise tracks of ETAMBOT 1 and ETAMBOT 2, and SABORD. The CO<sub>2</sub> flux is computed according to the parameterization of Wkf92. The CO<sub>2</sub> flux is positive when CO<sub>2</sub> escapes from the sea into the atmosphere and negative otherwise. Letters correspond to the different legs of the cruises (Fig. 1).

Table 1

CO<sub>2</sub> flux (mmol m<sup>-2</sup> day<sup>-1</sup>) computed according to the parameterizations of L&M86 and Wkf92 along each leg of ETAMBOT cruises and SABORD survey. Δ % represents the relative increase of CO<sub>2</sub> flux (Wkf92) vs. CO<sub>2</sub> flux (L&M86) (Δ % = 2(Wkf92 - L&M86)/(Wkf92 + L&M86))

	ETAMBOT 1 (September–October 1995)			ETAMBOT 2 (April – May 1996)		
	L&M86	Wkf92	Δ %	L&M86	Wkf92	Δ %
Cayenne — 50°W, 7.5°N (Leg A)	-0.21 ± 0.21	-1.17 ± 1.17	139	2.05 ± 1.28	3.32 ± 2.07	47
Section 7.5°N (50°W–35°W) (Leg B)	-0.22 ± 0.18	-0.69 ± 0.42	103	0.06 ± 1.03	0.11 ± 1.71	59
Section 35°W (7.5°N–5°S) (Leg C)	0.74 ± 0.48	1.16 ± 0.81	44	0.08 ± 0.66	0.40 ± 1.19	133
Section Ceara (0°, 45°W–8°N, 41°W) (Leg D)	0.04 ± 0.05	0.18 ± 0.22	127	0.95 ± 1.24	1.52 ± 1.99	46
50°W, 7.5°N — Cayenne (Leg E)				-11.53 ± 0.69	-18.68 ± 1.13	47
SABORD (May 1996)						
Continental shelf in front of French Guiana	L&M86	Wkf92	Δ %			
Mean (Legs A + B + C + D + E + F)	-15.0 ± 5.9	-24.2 ± 9.5	47			

tal shelf, is also related to the arrival of a freshwater lens originating from the Amazon mouth region and propagating northwestward along the coastline. The net CO<sub>2</sub> flux on the continental shelf is an order of magnitude (-24 mmol m<sup>-2</sup> day<sup>-1</sup>, after Wkf92) greater than the net CO<sub>2</sub> flux offshore (open sea areas) (Table 1).

### 5.3. Variability of the net CO<sub>2</sub> flux

Considering the strong regional relationship between oceanic *f*CO<sub>2</sub> and salinity (Fig. 5a), we infer the variability of the air–sea CO<sub>2</sub> flux in the western part of the equatorial belt, under the influence of the Amazon River outflow, from the variability of the SSS. Dessier and Donguy (1994) analyzed the seasonal and interannual variations of the SSS in the tropical Atlantic (15°S–30°N, 80°W–15°W) from observations collected by merchant ships between 1977 and 1989 (see in Fig. 7 the AX11 and AX20 WOCE XBT lines which indicate the track of the volunteer observing ships). They identified a well-defined seasonal cycle of SSS, related to river outflow (Amazon and Orinoco, northward at 8.5°N, 60.5°W) regimes. Near shore (5°N) the mean SSS cycle shows a minimum in May occurring just after the maximum Amazon outflow (Fig. 8a). Northeast of the shiptrack between Europe and French Guiana, the salinity minimum occurred in September–October near 10°N. This corresponds to the eastward

advection of the freshwater pool by the NECC. In the zonal belt 5°N–10°N (Fig. 8b), the low-salinity plume (*S* < 35) extends eastward as far as 35°W.

The air–sea flux of CO<sub>2</sub>, averaged over the whole area 5°S–10°N, 65°W–35°W, is calculated by bulk

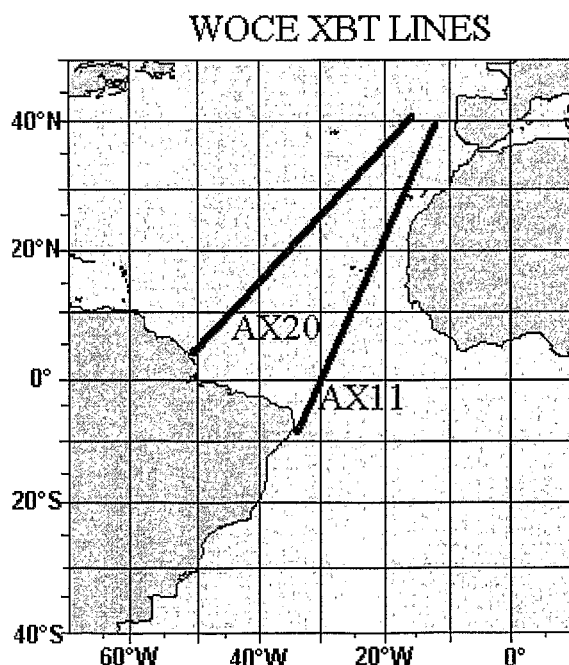
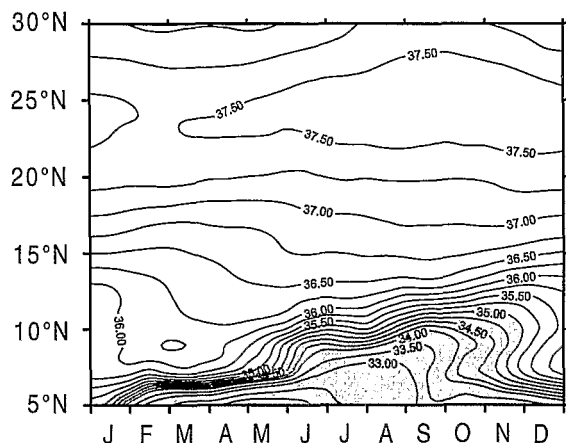


Fig. 7. Tracks of volunteer observing ships indicated by the WOCE XBT lines (AX20 between Europe and French Guiana and AX11 between Europe and Brazil).

## (a) Track Europe - French Guiana



## (b) Zonal belt 5°N-10°N

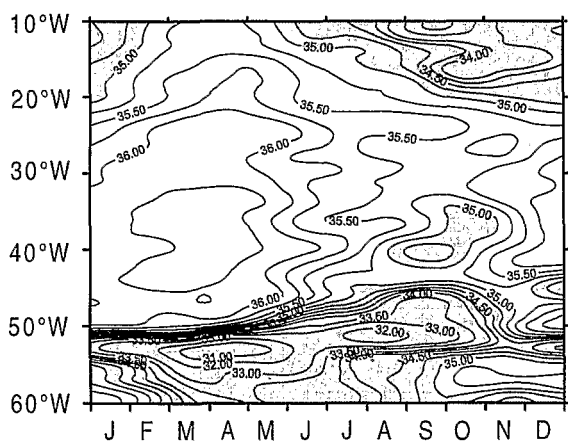


Fig. 8. Mean annual cycles of the sea surface salinity (space-time diagram) for the period 1977–1997, (a) along the shipping route from Europe to French Guiana and (b) in the zonal belt 5°N–10°N (from the South American coast to the African coast).

parameterization using climatologies of SST, wind velocity, salinity, and the  $f\text{CO}_2$  gradient between atmosphere and ocean. Within these latitudinal boundaries (5°S–10°N), the reduced mean annual variability of SST ( $< 2^\circ\text{C}$ ) could induce a limited effect on  $f\text{CO}_2$  variability ( $< 9\%$ , according to the  $\text{CO}_2$ -solubility temperature dependence of  $4.23\% \text{ } ^\circ\text{C}^{-1}$  for constant total  $\text{CO}_2$  of Takahashi et al., 1993) in comparison with  $f\text{CO}_2$  variability induced by variability of salinity ( $\sim 60\%$ , if we refer to the

range of variation of  $f\text{CO}_2$  at  $28^\circ\text{C}$ ,  $210\text{--}400 \mu\text{atm}$  in Fig. 4). Monthly climatological fields of SST and wind velocity have been extracted from the COADS monthly climatology<sup>1</sup>. SSS has been documented by updated long-term (1977–1997) monthly mean charts on a  $1^\circ \times 1^\circ$  grid constructed by objective analysis of all available data (bucket samples collected by merchant ships of the IRD — Institut de Recherche pour le Développement, previously Institut Français de Recherche pour le Développement en Coopération, ORSTOM — network completed with NODC World Ocean Atlas surface data for area without observations) (see Dessier and Donguy, 1994 for the kriging method). The oceanic  $f\text{CO}_2$  is determined from salinity using the linear relationship of Fig. 5a and the atmospheric  $f\text{CO}_2$  is chosen as the averaged value ( $353 \mu\text{atm}$ ) measured over the area in 1996 (ETAMBOT 2). The fields of monthly mean  $\text{CO}_2$  flux for April and September are shown in Fig. 9. As expected from SSS climatologies, the  $\text{CO}_2$  sink area is confined along the coastline in spring, while it expands seaward between 5°N and 10°N in summer, under the influence of the NBC retroflection. The fluxes estimated from cruises data (Fig. 6) are in relatively good agreement with these monthly mean flux fields. The escaping flux of  $2\text{--}3 \text{ mmol m}^{-2} \text{ day}^{-1}$  along  $7.5^\circ\text{N}$  (Leg B, ETAMBOT 2 in Fig. 6) is in the range ( $2\text{--}4 \text{ mmol m}^{-2} \text{ day}^{-1}$ ) of the April mean field of  $\text{CO}_2$  (Fig. 9). Nevertheless, this latter chart does not show the  $\text{CO}_2$  sink occurred at  $7.5^\circ\text{N}$ ,  $35^\circ\text{W}$  (end of Leg B and start of Leg C, Fig. 6, ETAMBOT 2). In September (Fig. 9, lower panel), the  $\text{CO}_2$  sink north of  $5^\circ\text{N}$ , which extends eastwards with the advection of freshwater by the NECC, is more pronounced ( $< -2 \text{ mmol m}^{-2} \text{ day}^{-1}$ ) than that estimated during ETAMBOT 1 ( $\sim -0.5 \text{ mmol m}^{-2} \text{ day}^{-1}$ , Fig. 6). On the continental shelf (Fig. 6, bottom), the range ( $-10$  to  $-40 \text{ mmol m}^{-2} \text{ day}^{-1}$ ) of  $\text{CO}_2$  flux measured during SABORD (May) is in good agreement with the lowest mean values indicated along the coast in April ( $< -20 \text{ mmol m}^{-2} \text{ day}^{-1}$ , Fig. 9). The monthly mean  $\text{CO}_2$  flux at the air–sea interface calculated according to both L&M86 and Wkf92 parameterizations over the area

<sup>1</sup> Server: <http://ferret.wrc.noaa.gov/fbin/climate>.



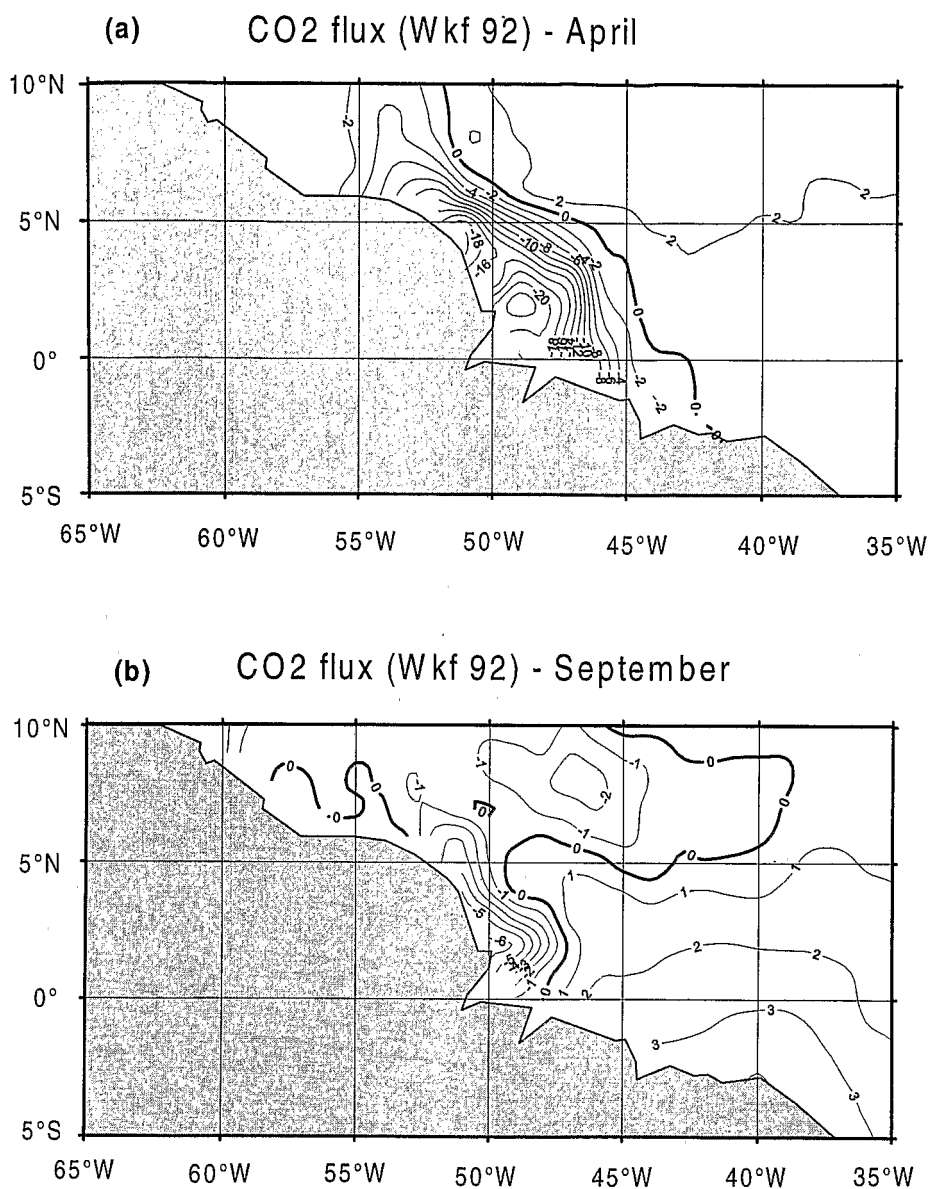


Fig. 9. Distribution of monthly mean CO<sub>2</sub> fluxes at the air–sea interface calculated from climatological fields of SST, wind velocity and SSS, according to the parameterization of Wkf92 (see text for the calculation method): (a) in April and (b) in September.

(5°S–10°N, 65°W–35°W) is shown in the Fig. 10d. On an annual average, the variabilities of monthly mean CO<sub>2</sub> flux and salinity are quite similar (Fig. 10b) and exhibit minima in April–June and maxima in December. The amplitude of the annual CO<sub>2</sub> flux variations reaches 2 mmol m<sup>-2</sup> day<sup>-1</sup>, according to the Wkf92 parameterization: the monthly CO<sub>2</sub> fluxes

range from -1.0 mmol m<sup>-2</sup> day<sup>-1</sup> in April to 1.0 mmol m<sup>-2</sup> day<sup>-1</sup> in December. The net annual mean CO<sub>2</sub> flux in the area 5°S–10°N, 65°W–35°W does not significantly differ from zero:  $0.08 \pm 0.37$  mmol m<sup>-2</sup> day<sup>-1</sup> using the L&M86 relationships and  $0.12 \pm 0.57$  mmol m<sup>-2</sup> day<sup>-1</sup> with the Wkf92 relationships. This result can be compared with the

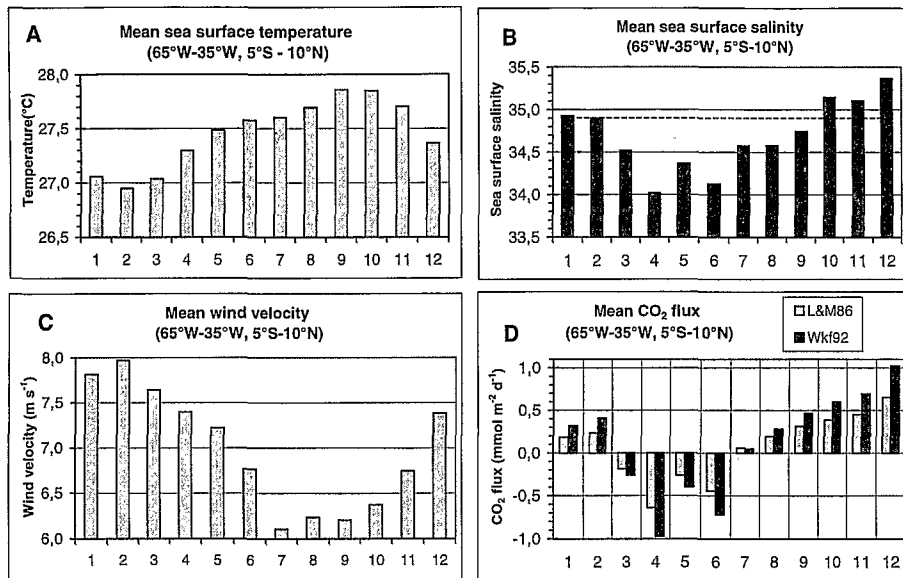


Fig. 10. Temporal diagrams of (A) monthly mean sea surface temperature, (B) monthly mean sea surface salinity, (C) monthly mean wind velocity, and (D) monthly mean CO<sub>2</sub> fluxes at the air–sea interface, over the area (5°S–10°N, 65°W–35°W). CO<sub>2</sub> fluxes are calculated from climatological fields (1977–1997) according to the parameterizations of L&M86 and Wkf92. The horizontal dashed line on the panel B indicates the salinity value (34.9) for which the oceanic CO<sub>2</sub> fugacity is equal to the atmospheric CO<sub>2</sub> fugacity (353 μatm) and the CO<sub>2</sub> flux goes to zero.

previous estimate of the annually mean CO<sub>2</sub> flux escaping from the Atlantic equatorial belt (5°S–5°N) between 35°W and 4°W from FOCAL cruises (Andrié et al., 1986) of 1.05 mmol m<sup>-2</sup> day<sup>-1</sup>, according to the L&M86 gas transfer–wind speed relationship. This clearly shows the role of river outflow (Amazon and secondarily Orinoco, the fourth most important river in the world) on CO<sub>2</sub> fluxes at the air–sea interface in the tropical Atlantic ocean. On an annual mean, the net source for atmospheric CO<sub>2</sub> observed in the central and eastern regions virtually disappears in the western part of the equatorial belt, where the *f*CO<sub>2</sub> level is lowered by rivers discharge.

## 6. Conclusions

The measurements of CO<sub>2</sub> fugacity at the air–sea interface in the western equatorial Atlantic ocean clearly shows that the area cannot be regarded as a source for atmospheric CO<sub>2</sub>, in contrast to the central and eastern parts of the equatorial belt. Rivers

run-off, primarily from the Amazon River, and secondarily from the Orinoco River, contribute to the decrease of oceanic *f*CO<sub>2</sub> by the dilution effect caused by the large input of freshwater. Fertilization of oceanic waters by river outflow enhances the biological pump of CO<sub>2</sub>, which contributes up to 30% to the lowering of *f*CO<sub>2</sub> in these areas. In spite of the large approximations, and of the uncertainties that remains in the CO<sub>2</sub> flux calculation, our study clearly highlights the role of freshwater inputs to the ocean in the global CO<sub>2</sub> budget.

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